# Practicing the Mole - - Even Problems Answer Key

Calculate the mass in grams of each of the following:

2. 8.00 moles of aluminum

6. 7.00 moles of iodine (I2)

4. 2.00 x 10<sup>2</sup> moles of chlorine (Cl<sub>2</sub>) 8. 9.20 moles of iron

Calculate the number of moles of atoms in each of the following:

10. 281 g of iron

14. 93.0 g of phosphorus

12. 72.0 g of magnesium

16. 1035 g of lead

Calculate the number of atoms in each of the following:

18. 5.40 moles of iron

20. 10.0 moles of titanium

# **Mole Practice Problem Answer Key**

**RJ Alexander** 

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