

# POLARITY REVIEW

1. Molarity is a type of \_\_\_\_\_, similar to  $\text{g/L}$ ,  $\text{mg/mL}$ ,  $\text{cm}^3/\text{cm}^3$ , etc. etc. etc.
2. Molarity is defined as  $M$  of \_\_\_\_\_ dissolved within one \_\_\_\_\_ of solution.
3.  $M$  is the unit for molarity, but the unit for molarity can also be written as:
4. Analyze the images below and identify the molarity of the solutions pictured.

2 moles  $\text{NaCl}$  dissolved in 2 liters of solution



\_\_\_\_\_ M

0.5 moles  $\text{NaCl}$  dissolved in 0.5 liters of solution



\_\_\_\_\_ M

4 moles  $\text{NaCl}$  dissolved in 4.0 liters of solution



\_\_\_\_\_ M

0.5 moles  $\text{NaCl}$  dissolved in 0.5 liters of solution



\_\_\_\_\_ M

2 moles  $\text{NaCl}$  dissolved in 0.25 liters of solution



\_\_\_\_\_ M

1 mole  $\text{NaCl}$  dissolved in 0.500 liters of solution



\_\_\_\_\_ M

5. Analyze the images below and identify how many total moles of the compound are present

1 L of 0.500 solution



\_\_\_\_\_ moles

1000 mL of 1.000 solution



\_\_\_\_\_ moles

2000 mL of 0.500 solution



\_\_\_\_\_ moles

0.5 L of 1.000 solution



\_\_\_\_\_ moles

2000 mL of 0.500 solution



\_\_\_\_\_ moles

0.25 L of 0.500 solution



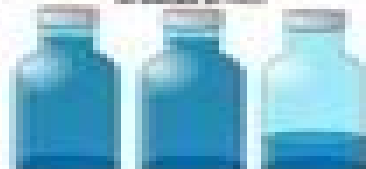
\_\_\_\_\_ moles

6. Analyze the images below and answer the prompts accordingly. Be sure to include the correct units and identity of the solution on question. Ex: 2 moles  $\text{CH}_4$  or 3M  $\text{MgO}$

How many moles of  $\text{NaCl}$  are in 1000 mL of a 0.500 solution?



What is the molar concentration of a solution containing 10 moles of  $\text{NaCl}$  in 2000 mL of solution?



How many moles of  $\text{NaCl}$  are in 0.5 L of a 0.500 solution?



Answer the below 3a and 3b with the number of grams. 3a. How many grams of sugar are in a 12 oz can of Coca-Cola? 3b. How many grams of sugar are in a 12 oz can of Diet Coke?



# Molarity Concept Review

**Charles Henrickson**



## **Molarity Concept Review:**

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