

Molar Mass

Worksheet-Answer Key

Name _____ Date _____

Exercise 1. How much does one mole weigh (molar mass/molecular weight) for the following? Show work: which numbers are added/multiplied, how many decimal places are used from the periodic table, and your result.

Na ___ 23g ___

22.99 → 23 g

NaF ___ 42g ___

23 + 19 = 42 g

Zn ___ 65.4 g ___

65.38 → 65.4 g

CO ___ 28 g ___

12 + 16 = 28 g

solid iron ___ 55.9 g ___

Fe 55.85 → 55.9 g

CO₂ ___ 44 g ___

12 + 2×16 = 12 + 32 = 44 g

Ca²⁺ (aq) ___ 40 g ___

same as Ca 40.08 → 40 g

CH₄ ___ 16 g ___

12 + 4×1 = 12 + 4 = 16

CaCO₃ ___ 100 g ___

40 + 12 + 3×16 = 100g

C₄H₁₀ ___ 58 g ___

4×12 + 10×1 = 58g

Molar Mass Practice With Answers

Michael Brown

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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