

Name \_\_\_\_\_ Class \_\_\_\_\_ Date \_\_\_\_\_

## Assessment

## Chapter Test A

## Chapter: Acids and Bases

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- \_\_\_\_\_ 1. Which of the properties listed below is *not* characteristic of an acid?
- a sour taste
  - a slippery feel
  - the ability to conduct an electric current
  - reactivity with metals
- \_\_\_\_\_ 2. Which is an example of a weak acid?
- HCl
  - $\text{H}_2\text{SO}_4$
  - $\text{H}_3\text{PO}_4$
  - $\text{HNO}_3$
- \_\_\_\_\_ 3. Acetic acid is found in
- vinegar.
  - the stomach.
  - antacids.
  - oranges.
- \_\_\_\_\_ 4. A characteristic of an Arrhenius base is that it
- is an electrolyte.
  - tastes bitter.
  - dissociates to form  $\text{OH}^-$  ions.
  - All of the above
- \_\_\_\_\_ 5. Which of the following is a Brønsted-Lowry base?
- HCl
  - $\text{HCO}_3^-$
  - $\text{H}_3\text{O}^+$
  - $\text{H}_3\text{PO}_4$
- \_\_\_\_\_ 6. In the reaction represented by the equation
- $$\text{H}_2\text{C}_2\text{O}_4(aq) + \text{CH}_3\text{NH}_2(aq) \rightleftharpoons \text{HC}_2\text{O}_4^-(aq) + \text{CH}_3\text{NH}_3^+(aq),$$
- which of these is a conjugate acid-base pair?
- $\text{H}_2\text{C}_2\text{O}_4(aq)$  and  $\text{CH}_3\text{NH}_2(aq)$
  - $\text{H}_2\text{C}_2\text{O}_4(aq)$  and  $\text{CH}_3\text{NH}_3^+(aq)$
  - $\text{CH}_3\text{NH}_2(aq)$  and  $\text{CH}_3\text{NH}_3^+(aq)$
  - $\text{HC}_2\text{O}_4^-(aq)$  and  $\text{CH}_3\text{NH}_3^+(aq)$

# Modern Chemistry Test A Chapter 5 Answers

**TD Snyder**



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are these chemical and physical structures experimentally determined The existing methods for polymer characterization are enumerated and discussed in this chapter The following chapters go into more detail For most applications textiles films molded or extruded objects of all kinds the mechanical and the thermal behaviors of polymers are of pre ponderant importance followed by optical and electric properties Chapters 4 through 9 describe how such properties are rooted in and dependent on the chemical structure More detailed considerations are given to certain particularly important and critical properties such as the solubility and permeability of polymeric systems Macromolecules are not always the final goal of the chemist they may act as intermediates reactants or catalysts This topic is presented in Chapters 10 and 11

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