

## Chemical Equations and Reactions

### SECTION 2

**SHORT ANSWER** Answer the following questions in the space provided.

1. Match the equation type on the left to its representation on the right.

\_\_\_ **c** \_\_\_ synthesis



\_\_\_ **d** \_\_\_ decomposition



\_\_\_ **b** \_\_\_ single-displacement



\_\_\_ **a** \_\_\_ double-displacement



2. \_\_\_ **c** \_\_\_ In the reaction described by the equation  $2Al(s) + 3Fe(NO_3)_2(aq) \rightarrow 3Fe(s) + 2Al(NO_3)_3(aq)$ , iron has been replaced by

(a) nitrate.

(c) **aluminum.**

(b) water.

(d) nitrogen.

3. \_\_\_ **a** \_\_\_ Of the following chemical equations, the only reaction that is both synthesis and combustion is



4. \_\_\_ **b** \_\_\_ Of the following chemical equations, the only reaction that is both combustion and decomposition is



(d)

5. Identify the products when the following substances decompose:

a. **Its separate elements** \_\_\_\_\_ a binary compound

b. **Metal oxide + water** \_\_\_\_\_ most metal hydroxides

c. **Metal oxide + carbon dioxide** \_\_\_\_\_ a metal carbonate

d. **Water + sulfur dioxide** \_\_\_\_\_ the acid  $H_2SO_3$

6. The complete combustion of a hydrocarbon in excess oxygen yields the products

# Modern Chemistry Chapter 8 Section 2 Review Answers

**Leo J. Malone**



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