Chapter 3: Matter and Energy - Study Guide

*Do not forget about the extra resources available to help you study and prepare for this test!

- Textbook pages 81-93 (answers to odd questions in the back of the book)
- Mastering Chemistry Study Area: http://www.pearsonmylabandmastering.com/northamerica/
- Miss Marnik's Website: http://www.northallegheny.org/Page/16665
- Labs, problem packets, and homework assignments

Section 1 - Classifying Matter: (text pages 57-62)

Need to be able to:

- Define matter, atoms, molecules, element, compound, and mixture.
- Classify matter as and distinguish between solid, liquid, or gas.
- Classify matter as element, compound, or mixture.

Practice Problems:

- Match the following terms with the phrases below:
 - State of matter in which atoms or molecules pack close to each other in fixed locations
 - State of matter in which atoms or molecules are packed close together but are free to move around and by each other
 - c. Substance composed of only one type of atom or molecule
 - Matter composed of two or more different types of atoms or molecules combined in variable proportions
 - Mixture that has two or more regions with different compositions
 - f. Mixture that has the same composition throughout
- 2. Which of the following can be classified as matter?
 - a. atoms c. wind

e. potential energy

b. gas

d. liquids

f. temperature

g. carbon h. light

heterogeneous mixture

homogeneous mixture

liquid

solid

mixture.

pure substance.

- Classify each type of matter as a pure substance or a mixture. If it is a pure substance, classify it as an
 element or a compound. If it is a mixture, classify it as homogeneous or heterogeneous.
 - a. limestone (CaCo₁)

e. Chex mix

b. orange juice with pulp

f. Table salt

c. helium

g. Brass

d. Pacific ocean

h. oxygen

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Table of Contents Matter And Energy Study Guide

- 1. Understanding the eBook Matter And Energy Study Guide
 - The Rise of Digital Reading Matter And Energy Study Guide
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Matter And Energy Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Matter And Energy Study Guide
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Matter And Energy Study Guide
 - Personalized Recommendations
 - Matter And Energy Study Guide User Reviews and Ratings
 - Matter And Energy Study Guide and Bestseller Lists
- 5. Accessing Matter And Energy Study Guide Free and Paid eBooks

- Matter And Energy Study Guide Public Domain eBooks
- Matter And Energy Study Guide eBook Subscription Services
- Matter And Energy Study Guide Budget-Friendly Options
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 - Matter And Energy Study Guide Compatibility with Devices
 - Matter And Energy Study Guide Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Matter And Energy Study Guide
 - Highlighting and Note-Taking Matter And Energy Study Guide
 - Interactive Elements Matter And Energy Study Guide
- 8. Staying Engaged with Matter And Energy Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Matter And Energy Study Guide
- 9. Balancing eBooks and Physical Books Matter And Energy Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Matter And Energy Study Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Matter And Energy Study Guide
 - Setting Reading Goals Matter And Energy Study Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Matter And Energy Study Guide
 - Fact-Checking eBook Content of Matter And Energy Study Guide
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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