

(c) Standard enthalpy changes of reaction can also be calculated using mean bond enthalpies.

(i) What is meant by the term **mean bond enthalpy**?

(2)

(ii) Describe the bonding in a $\text{C}=\text{C}$ double bond in terms of the different ways in which the orbitals overlap.

You may draw a diagram if you wish.

(2)

Space for diagram:



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