

# TITRATIONS



Calculate the concentration of the sulfuric acid in mol/dm<sup>3</sup>

Use only the student's concordant results

mean titre =  $\frac{22.10 + 22.15 + 22.15}{3} = 22.13 \text{ cm}^3$  ✓

$$n(\text{NaOH}) = 0.105 \times (22.13/1000) = 0.002324 \text{ mol} \quad \checkmark$$

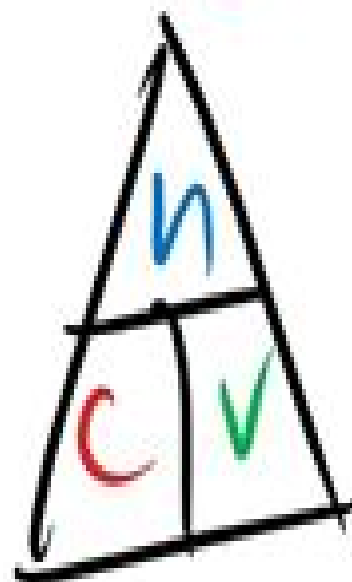
$$n(\text{H}_2\text{SO}_4) = 0.002324 / 2 = 0.001162 \text{ mol} \quad \checkmark$$

$$c(\text{H}_2\text{SO}_4) = \frac{0.001162}{(25/1000)} = 0.0465 \text{ mol/dm}^3 \quad \checkmark$$

$$n = m/M_r$$

$$n = c \times v$$

$$c = n/v$$



# GCSE CHEMISTRY

# Gcse Titration Practise Questions

**L Manion**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

## Gcse Titration Practise Questions:

## Unveiling the Magic of Words: A Report on "**Gcse Titration Practise Questions**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Gcse Titration Practise Questions**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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