

# Periodic Table Basics

**Step 1:** Complete the square for each element by filling in the atomic number, name, & atomic mass.

**Step 2:** Determine the number of protons, neutrons, and electrons in an atom of each element.

**Step 3:** Identify if the element is a solid, liquid, or gas at room temperature.

**Step 4:** Give the melting (M.P.) and boiling points (B.P.) in degrees Celsius.

**Step 5:** List at least three physical or chemical properties for each element.

**Step 6:** List at least three uses for each element.

**Step 7:** Draw a Bohr diagram and Lewis Structure to show the arrangement of electrons and the number of valence electrons.

**Step 8:** Use the following colors to shade in the square for each element. You should **ONLY** color the small square in the upper left-hand corner and not the entire card.

Green = Li & Na  
Orange = B & Al

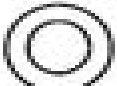
Pink = O & S  
Red = C & Si

Blue = Be & Mg  
Tan = N & P

Purple = F & Cl  
Yellow = He, Ne, & Ar

**Step 9:** Cut the cards apart and arrange according to atomic number in the pattern shown. Once you have the cards arranged in the correct order, glue them to a large sheet of construction paper.

1							2
3	4	5	6	7	8	9	10
11	12	13	14	15	16	17	18

① <div style="text-align: center; font-size: 2em;">B</div>  	P= ② N= E=	③ O S O L O G
	M.P. = B.P. = ④	
Properties ⑤		
Uses ⑥		
Bohr Diagram      Lewis Structure		
<div style="display: flex; align-items: center; justify-content: space-around;">  <div style="text-align: center;">         ⑦   <div style="font-size: 1.5em;">B</div> </div> </div>		

Done? Answer the questions on the worksheet using the information on your Periodic Table!

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**Page 87 If8767**

**JR Anderson**



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## **Table of Contents Light Matching Worksheet Answers Instructional Fair Page 87 If8767**

1. Understanding the eBook Light Matching Worksheet Answers Instructional Fair Page 87 If8767
  - The Rise of Digital Reading Light Matching Worksheet Answers Instructional Fair Page 87 If8767
  - Advantages of eBooks Over Traditional Books
2. Identifying Light Matching Worksheet Answers Instructional Fair Page 87 If8767
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Light Matching Worksheet Answers Instructional Fair Page 87 If8767
  - User-Friendly Interface
4. Exploring eBook Recommendations from Light Matching Worksheet Answers Instructional Fair Page 87 If8767
  - Personalized Recommendations
  - Light Matching Worksheet Answers Instructional Fair Page 87 If8767 User Reviews and Ratings
  - Light Matching Worksheet Answers Instructional Fair Page 87 If8767 and Bestseller Lists
5. Accessing Light Matching Worksheet Answers Instructional Fair Page 87 If8767 Free and Paid eBooks

- Light Matching Worksheet Answers Instructional Fair Page 87 If8767 Public Domain eBooks
  - Light Matching Worksheet Answers Instructional Fair Page 87 If8767 eBook Subscription Services
  - Light Matching Worksheet Answers Instructional Fair Page 87 If8767 Budget-Friendly Options
6. Navigating Light Matching Worksheet Answers Instructional Fair Page 87 If8767 eBook Formats
- ePub, PDF, MOBI, and More
  - Light Matching Worksheet Answers Instructional Fair Page 87 If8767 Compatibility with Devices
  - Light Matching Worksheet Answers Instructional Fair Page 87 If8767 Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Light Matching Worksheet Answers Instructional Fair Page 87 If8767
  - Highlighting and Note-Taking Light Matching Worksheet Answers Instructional Fair Page 87 If8767
  - Interactive Elements Light Matching Worksheet Answers Instructional Fair Page 87 If8767
8. Staying Engaged with Light Matching Worksheet Answers Instructional Fair Page 87 If8767
- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Light Matching Worksheet Answers Instructional Fair Page 87 If8767
9. Balancing eBooks and Physical Books Light Matching Worksheet Answers Instructional Fair Page 87 If8767
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Light Matching Worksheet Answers Instructional Fair Page 87 If8767
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Light Matching Worksheet Answers Instructional Fair Page 87 If8767
- Setting Reading Goals Light Matching Worksheet Answers Instructional Fair Page 87 If8767
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Light Matching Worksheet Answers Instructional Fair Page 87 If8767
- Fact-Checking eBook Content of Light Matching Worksheet Answers Instructional Fair Page 87 If8767
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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