

# 9 • Bonding & Molecular Structure

## PRACTICE TEST

1. The correct Lewis symbol for ground state carbon

is (a)  $\cdot\dot{C}\cdot$  b)  $\cdot\dot{C}\cdot$  c)  $\cdot\dot{C}\cdot$  d)  $\cdot\dot{C}\cdot$  e)  $\cdot\dot{C}\cdot$

*Handwritten notes: "FOUR BONDS" with arrows pointing to the four dots in option (a). "A.L." with an arrow pointing to the central C atom.*

2. The correct Lewis symbol for ground state aluminum is

a)  $\cdot\dot{Al}\cdot$  b)  $\cdot\dot{Al}\cdot$  (c)  $\cdot\dot{Al}\cdot$  d)  $\cdot\dot{Al}\cdot$  e)  $\cdot\dot{Al}\cdot$

3. Using the picture below, what process corresponds to the lattice energy?

$Li(g) + F(g) \rightarrow LiF(s)$

a) 1 b) 2 c) 3 d) 4 (e) 5

*Handwritten notes: "Li(g) + F(g) -> LiF(s)" written below the diagram. A red circle highlights the final state of the diagram.*

4. Which of the following favors formation of an ionic compound?

- (a) low ionization energy for metal
- (b) high electron affinity for non metal
- (c) high lattice energy
- (d) all of a-c above
- e) none of a-c above

5. Which of the atoms below is least likely to violate the octet rule?

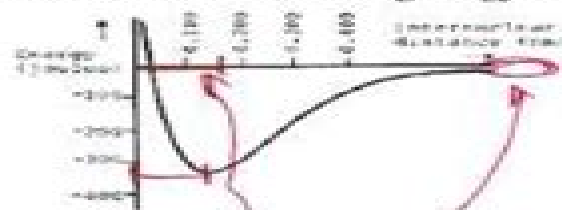
- a) Be b) P c) S d) B (e) F
- Handwritten notes: "FAM IF" under Be, "CAN DO EXTENDED VAL" under P, "FAM II" under B.*

6. How many electrons are shown in the Lewis structure of perchlorate ion,  $ClO_4^-$ ?

- a) 30 b) 31 (c) 32 d) 50 e) 51



Questions 7 - 9 refer to the following energy diagram:



7. What is the energy of the two isolated atoms?

- a) -400 J d) 0.100 nm
- b) -335 J e) -155 J
- (c) 0 J

8. What is the bond length of the bond between the two atoms?

- a) 0.020 nm d) 0.330 nm
- (b) 0.140 nm e) 2.00 nm
- c) 0.400 nm

9. What is the bond energy (bond strength) of the bond?

- a) 0 J (d) -330 J
  - b) -110 J e) -422 J
  - c) -10 J
- Handwritten notes: "SHOULD THIS BE (+) VALUE?" and "Minimum" with an arrow pointing to the -330 J option.*

10. What is the bond order of the C-O bond in acetone?

$CH_3-C(=O)-CH_3$

a) 4 b) 1.5 c) 0.5 d) 1 (c) 2

*Handwritten note: "double = B.O 2" with an arrow pointing to the double bond in the acetone structure.*

# Larson Geometry Chapter Two Test Answers

**M Walker**



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