

## Atomic History Worksheet Answer Key

### Democritus Proposes the Atom

1. The word *atom* comes from a Greek word that means **indivisible**.
2. Which of the following statements is part of Democritus's theory about atoms?
  - a. **Atoms are the smallest piece of matter.**
  - d. **Atoms exist but will never be fully proven.**
3. We know that Democritus was right about atoms. So why did people ignore Democritus's ideas for such a long time?  
**Democritus's ideas were ignored for a long time because he had no proof of the atoms existence and people believed Aristotle's ideas.**

### Dalton Creates an Atomic Theory Based on Experiments

4. By conducting experiments and making observations, Dalton figured out that elements combine in random proportions because they're made of individual atoms. **False**
5. Dalton's theory states that atoms cannot be **changed, created, or destroyed**.
6. Atoms of different elements are exactly alike. **False**
7. How did Dalton think atoms formed new substances?  
**Atoms can be combined in set proportions to form new substances.**

### Thomson Finds Electrons in the Atom

Mark the following statements *True* or *False*.

8. In 1897, J. J. Thomson made a discovery that proved Dalton's atomic model was correct. **False**
9. Thomson discovered that there were small particles inside the atom. **True**
10. Thomson found that the electrically charged plates affected the direction of a cathode-ray tube beam. **True**
11. Thomson knew the beam was made of particles with a positive charge because it was pulled toward a positive charge. **False**
12. The two types of charges are positive and neutral. **False**

# History Of The Atom Answer Key

**RC Schank**



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