

19.2

HYDROGEN IONS AND ACIDITY

Section Review

Objectives

1. Identify a conjugate acid-base pair in chemical reactions involving acids and bases.
2. Calculate the pH of a solution from the concentration of H^+ or H_3O^+ ions, and vice versa.
3. Determine the percent ionization of a weak acid.

Vocabulary

acid	1. conjugate base	acid ionization
base	2. hydronium	acid ionization constant
base ionization	3. autoionization	pH

Key Equation

- a. $pH = -\log[H^+]$ (eq. 19.2)
- b. $[H^+] = 10^{-pH}$ (eq. 19.3)

Part A Completion

Write the chemical formula or identify the substance in each space, and then name the substance. Write the chemical formula in the space, and then name the substance.

1. The chemical formula of the conjugate base of HNO_3 is _____.
2. The chemical formula of the conjugate acid of OH^- is _____.
3. The chemical formula of the conjugate base of H_2SO_4 is _____.
4. The chemical formula of the conjugate acid of CO_3^{2-} is _____.
5. The chemical formula of the conjugate base of $H_2PO_4^-$ is _____.
6. The chemical formula of the conjugate acid of HSO_4^- is _____.
7. The chemical formula of the conjugate base of H_2O is _____.
8. The chemical formula of the conjugate acid of H_2O is _____.
9. The chemical formula of the conjugate base of H_2O is _____.
10. The chemical formula of the conjugate acid of H_2O is _____.
11. The chemical formula of the conjugate base of H_2O is _____.
12. The chemical formula of the conjugate acid of H_2O is _____.
13. The chemical formula of the conjugate base of H_2O is _____.
14. The chemical formula of the conjugate acid of H_2O is _____.
15. The chemical formula of the conjugate base of H_2O is _____.
16. The chemical formula of the conjugate acid of H_2O is _____.
17. The chemical formula of the conjugate base of H_2O is _____.
18. The chemical formula of the conjugate acid of H_2O is _____.
19. The chemical formula of the conjugate base of H_2O is _____.
20. The chemical formula of the conjugate acid of H_2O is _____.

Hydrogen Ions And Acidity Section Review Answers

RC Schank



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