

- The entropy change of the surroundings is driven by heat flow.
 - 2nd law: every energy transfer or transformation increases the entropy of the universe.
- c. Conservation of matter in chemical systems.
- Matter neither created nor destroyed, so we must have the same number and type of atoms after the chemical change as were present before the chemical change.
- d. Kinetic and potential energy
- **Kinetic energy:** energy in motion.
 - **Potential energy:** the energy possessed by a body by virtue of its position relative to others, stresses within itself, electric charge, and other factors.
- e. Transformations between different forms of energy (thermal, chemical, radiant, nuclear, mechanical, electrical, electromagnetic, sound, and chemical).
- **Energy Transformations Examples:** An energy transformation is the change of energy from one form to another. Energy transformations occur everywhere every second of the day. There are many different forms of energy such as electrical, thermal, nuclear, mechanical, electromagnetic, sound, and chemical.

f. Differences between chemical and physical properties/changes

- The difference between a physical reaction and a chemical reaction is composition.
- **Chemical property/changes:** The changing of matter of a substance is not necessarily an indication of a chemical change. For example, changing the color of a metal does not change its physical properties; however, in a chemical reaction, a color change is usually an indication that a reaction is occurring.
 - Change in Temperature
 - Change in Color
 - Noticeable Odor (after reaction has begun)
 - Formation of a Precipitate
 - Formation of Bubbles
- **Physical property/changes:** A physical change there is a difference in the appearance, smell, or simple display of a sample of matter without a change in composition.
 - Texture
 - Color
 - Temperature
 - Shape
 - Change of State (Boiling Point and Melting Point are significant factors in determining this change.)

Physical properties include many other aspects of a substance. The following are (but not limited to) physical properties:

- Luster
- Malleability
- Ability to be drawn into a thin wire
- Density
- Viscosity
- Solubility
- Mass
- Volume

g. Various temperature scales (Celsius, Fahrenheit, Kelvin)

General Science Praxis Study Guide

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Praxis II General Science Content Knowledge 5435 Study Guide Praxis 5435 Study Guide Team, 2017-05-18 Praxis II General Science Content Knowledge 5435 Study Guide Test Prep Practice Test Questions for the Praxis 2 General Science Exam Developed for test takers trying to achieve a passing score on the Praxis II General Science Exam this comprehensive study guide includes Quick Overview Test Taking Strategies Introduction Scientific Methodology Techniques and History Physical Science Life Science Earth and Space Science Science Technology and Society Practice Questions Detailed Answer Explanations Each section of the test has a comprehensive review that goes into detail to cover all of the content likely to appear on the Praxis II General Science Exam The practice test questions are each followed by detailed answer explanations If you miss a question it is important that you are able to understand the nature of your mistake and how to avoid making it again in the future The answer explanations will help you to learn from your mistakes and overcome them Understanding the latest test taking strategies is essential to preparing you for what you will expect on the exam A test taker has to not only understand the material that is being covered on the test but also must be familiar with the strategies that are necessary to properly utilize the time provided and get through the test without making any avoidable errors Anyone planning to take the Praxis II General Science Exam should take advantage of the review material practice test questions and test taking strategies contained in this study guide

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