

Elements and Their Properties Test**Multiple Choice**

Choose the choice that best completes the statement or answers the question.

Testing Concepts

- _____ 1. An allotrope of carbon that is soft and can be used as a lubricant is _____.
a. diamond
b. charcoal
c. coal
d. graphite
- _____ 2. The process by which a solid changes directly to a gas without first becoming a liquid is called _____.
a. condensation
b. ionization
c. sublimation
d. evaporation
- _____ 3. In the gaseous state, _____ form diatomic covalent molecules.
a. noble gases
b. alkali metals
c. alkaline earth metals
d. halogens
- _____ 4. When hydrogen reacts with the active metals, it _____.
a. shares electrons
b. gains one electron
c. loses one electron
d. loses two electrons
- _____ 5. Different forms of the same element that have different properties because of different atom arrangements are called _____.
a. isotopes
b. carbon
c. diamond
d. graphite
- _____ 6. At room temperature, most metals are _____.
a. liquids
b. solids
c. radioactive
d. gases
- _____ 7. An allotrope of carbon that is hard and is referred to as jewelry is _____.
a. coal
b. charcoal
c. quartz
d. graphite
- _____ 8. Metals can be used as wires because they are _____.
a. ductile
b. malleable
c. shiny
d. alloys
- _____ 9. The _____ often occur in nature as uncombined elements.
a. alkaline earth metals
b. alkali metals
c. noble gases
d. transition elements
- _____ 10. Substances that conduct an electric current only under certain conditions are most likely to be _____.
a. metals
b. semimetals
c. metalloids
d. noble gases

Elements And Their Properties Test Answer

J Elliott



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