

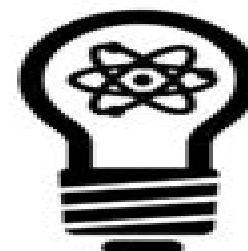
ATOMS AND PERIODIC TABLE QUIZ

Do Not Write on This Test!!!! You may use a periodic table.

Form B

Choose the BEST answer.

- The zigzag line on the periodic table divides
 - alkali metals and transition metals
 - metals and nonmetals
 - semimetals and transition metals
 - inert gases and halogens
- The smallest particle into which an element can be divided and still be the same substance is a (n)
 - electron
 - proton
 - atom
 - molecule
- Electrons are particles that
 - help make up the nucleus and have no charge
 - help make up the nucleus and are positively charged
 - are located outside of the nucleus and are negatively charged
 - are located outside of the nucleus and have no charge
- If you found a Carbon 13 atom, you would know that
 - it has 13 protons
 - it has 13 electrons
 - it has 13 neutrons
 - it has 7 neutrons
- What particles make up the nucleus?
 - electrons and neutrons
 - electrons and protons
 - protons and neutrons
 - electrons, protons, and neutrons
- Neutrons are particles that
 - help make up the nucleus and have no charge
 - help make up the nucleus and are positively charged
 - are located outside of the nucleus and are negatively charged
 - are located outside of the nucleus and have no charge
- The current model of the atom suggests that
 - proton clouds surround the nucleus
 - electron clouds surround the nucleus
 - electrons travel in definite paths around the nucleus
 - the exact path of a moving electron can be predicted
- The mass number minus the atomic number is equal to the number of
 - protons and neutrons
 - protons
 - protons and electrons
 - neutrons
- Mendeleev arranged the elements
 - density
 - melting point
 - how they look (appearance)
 - atomic mass
- Which of the following is NOT found on the periodic table?
 - the atomic number of each element
 - the symbol of each element
 - the density of each element
 - the atomic mass of each element



Elements And The Periodic Table Chapter Test

**Ralph L. Shriner, Christine K. F.
Hermann, Terence C. Morrill, David Y.
Curtin, Reynold C. Fuson**

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