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Electrons in Atoms Study Worksheet

Fill in the blanks in the fo	ollowing paragraph with	n the appropriate terms.
In Bohr's model of the at	om, electrons inhabit ce	ertain levels,
with the levels closest to	o the nucleus having	energy than those
farther from the nucleus	. In the st	tate of the atom, the electrons
are in the lowest	level possible. W	hen an atom absorbs energy,
it is said to ascend to an	state, wh	ere it is unstable. The atom
will soon the	same amount of energy	absorbed which may be seen
in the form of visible ligh	nt. In the study of	, this visible ligh
is seen as the	spectrum o	of an element. This spectrum is
to each e	lement, hence it is also	called an element's
"fingerprints". The mode	rn view of light is that it	has a nature.In
other words, light may b	ehave as a stream of par	rticles called or
, or light	may behave as a	
In the wave view of light	, the wave equation is of	ften used to determine a
wave's frequency or wavelength. The		is the distance
between corresponding	points on adjacent wave	es while the
is the number of waves p	passing a given point in a	a given time. The wave
equation is:		

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