

Electron Energy and Light

How does light reveal the behavior of electrons in an atom?

Why?

From fireworks to stars, the color of light is useful in finding out what's in matter. The emission of light by hydrogen and other atoms has played a key role in understanding the electronic structure of atoms. Trace materials, such as evidence from a crime scene, lead in paint or mercury in drinking water, can be identified by heating or burning the materials and examining the color(s) of light given off in the form of bright-line spectra.

Model 1 – White Light



- Trace the arrows in Model 1 and shade in the table with colored pencils where appropriate.
- What happens to white light when it passes through a prism?

The different wavelengths get separated

- Why are the color labels in the table in Model 1 plural (i.e., "Reds" rather than "Red")?

many different shades of a color in a range of wavelength

- Do all colors of light travel at the same speed?

Yes

- Do all colors of light have the same energy? If no, which colors have the highest energy and the least energy, respectively?

*All colors do not have the same energy. Blues + violets highest
Reds + oranges lowest*

- Consider the light illustrated in Model 1.

- Which color corresponds to the longest wavelengths?

Reds

- Which color corresponds to the shortest wavelengths?

Violets

- Write a sentence that describes the relationship between wavelength and energy of light.

As the wavelength gets shorter the energy of light increases

Electron Energy And Light Key

**Nicholas J. Heller, Audrey J.
Washington, Scott K. Cushing**



Electron Energy And Light Key:

Electron Energy Loss Spectroscopy Nicholas J. Heller, Audrey J. Washington, Scott K. Cushing, 2025-06-12 This primer is geared toward undergraduate and graduate students who possess a basic understanding of chemistry and physics and are looking for a route to enter the world of electron energy loss spectroscopy EELS It may also be a convenient reference for established researchers interested in adopting a new analytical method or simply for anyone eager to learn more about electron matter interactions This primer is divided into five chapters to guide your reading Chapter 1 introduces the concept of electron based spectroscopy and defines what qualifies as an electron based technique The authors outline the principal interactions at play and present a range of methods that correspond to different types of signals The chapter concludes with a concise historical narrative tracing the development of EELS Chapter 2 details the background required to build a deeper understanding of EELS and spectroscopy The authors connect photon based and electron based approaches highlighting where they overlap and where they diverge Chapter 3 focuses on the fundamental instrumentation used in electron spectroscopy discusses optimization strategies and introduces key instrument parameters The authors compare different electron sources optical geometries and lens configurations Chapter 4 explores scattering mechanisms and electron matter interactions offering theoretical underpinnings and guidance on interpreting EELS spectra Key concepts such as elastic and inelastic scattering core level transitions plasmons and analysis methods are reviewed Chapter 5 concludes with real world applications such as the use of EELS in biology nanomaterials polymers and solid state systems The authors also briefly survey emerging directions including in situ and ultrafast EELS underscoring the method s evolving landscape

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