

## Substituted Molecular Compounds

Obtain the following using Substituted Compounds

Take great care that Compounds that are not Simple Ions

Number

In these compounds, which are formed between simple metals and nonmetals, the atoms bond together to form molecules (not, as we have

compounds, substances and used to show the number of atoms of each element per molecule. Note the subscript, giving the prefix and the number of atoms each represents:

|   |      |   |    |   |     |   |       |   |       |   |      |   |       |   |      |   |      |    |      |
|---|------|---|----|---|-----|---|-------|---|-------|---|------|---|-------|---|------|---|------|----|------|
| 1 | mono | 2 | di | 3 | tri | 4 | tetra | 5 | penta | 6 | hexa | 7 | hepta | 8 | octa | 9 | nona | 10 | deca |
| 1 |      | 2 |    | 3 |     | 4 |       | 5 |       | 6 |      | 7 |       | 8 |      | 9 |      | 10 |      |

Write the formula of these substituted compounds:

carbon hydrosulfide  $\text{CSH}_4$

nitric oxide  $\text{NO}$

nitrogen monoxide  $\text{N}_2\text{O}$

nitrogen trioxide  $\text{N}_2\text{O}_3$

nitrogen pentoxide  $\text{N}_2\text{O}_5$

nitric oxide  $\text{NO}$

phosphorus trichloride  $\text{PCl}_3$

carbon dioxide  $\text{CO}_2$

orthophosphoric acid  $\text{H}_3\text{PO}_4$

carbon monoxide  $\text{CO}$

## Complex Compounds, a Special Case of Substituted Compounds

Organic compounds are more difficult to compound than inorganic compounds that are formed.

Carbon

The nature of the simple molecules that give off

carbon atoms is not organic compound and is therefore different from compound in which it is substituted. This simple substance that contains carbon is

methane compound and simple gases that are of methane compound

that simple substance is hydrocarbon and is simple

simple gases. The organic compounds, containing only hydrogen and

oxygen is hydrocarbon

# Gpb Chemistry Episode 8answers

**Shasha Hu**



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