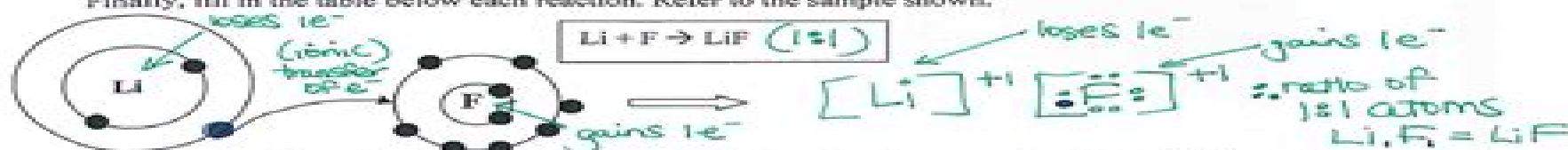


Name Solutions

Date _____ Period _____

Ionic Bonding Worksheet

For each pair of elements below draw an atomic diagram showing electrons in different energy levels. Draw arrows to show where the outer electrons will go during a chemical reaction, then draw the resulting compound. Finally, fill in the table below each reaction. Refer to the sample shown.



Atoms	Valence electrons	Electron transfer from/to each atom	Ions formed in the product
Li	1 ve	transfers $1e^-$ to F	Li^+
F	7 ve	accepts $1e^-$ from Li	F^-

Reactions	Atoms	Valence electrons	Electron transfer from/to each atom	Ions formed in the product
1) $Li + Cl \Rightarrow LiCl$	Li	1 ve	donates $1e^-$	Li^+
$Li + Cl \Rightarrow LiCl$	Cl	1 ve	accepts $1e^-$	Cl^-
2) $Ca + O \Rightarrow CaO$	Ca	2 ve	donates $2e^-$	Ca^{+2}
$Ca + O \Rightarrow CaO$	O	6 ve	accepts $2e^-$	O^{2-}
3) $Be + F \Rightarrow BeF_2$	Be	2 ve	donates $2e^-$ to each F	Be^{+2}
$Be + F \Rightarrow BeF_2$	F	7 ve	each F accepts $1e^-$ each	F^-
4) $Mg + S \Rightarrow MgS$	Mg	2 ve	donates $2e^-$ to S	Mg^{+2}
$Mg + S \Rightarrow MgS$	S	6 ve	accepts $2e^-$	S^{2-}
5) $K + F \Rightarrow KF$	K	1 ve	donates $1e^-$	K^+
$K + F \Rightarrow KF$	F	7 ve	accepts $1e^-$	F^-

Elements And Bonding Worksheet Answers

RS Peters

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the 'RS Peters' text.

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Table of Contents Elements And Bonding Worksheet Answers

1. Understanding the eBook Elements And Bonding Worksheet Answers
 - The Rise of Digital Reading Elements And Bonding Worksheet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Elements And Bonding Worksheet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Elements And Bonding Worksheet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Elements And Bonding Worksheet Answers

- Personalized Recommendations
- Elements And Bonding Worksheet Answers User Reviews and Ratings
- Elements And Bonding Worksheet Answers and Bestseller Lists
- 5. Accessing Elements And Bonding Worksheet Answers Free and Paid eBooks
 - Elements And Bonding Worksheet Answers Public Domain eBooks
 - Elements And Bonding Worksheet Answers eBook Subscription Services
 - Elements And Bonding Worksheet Answers Budget-Friendly Options
- 6. Navigating Elements And Bonding Worksheet Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Elements And Bonding Worksheet Answers Compatibility with Devices
 - Elements And Bonding Worksheet Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Elements And Bonding Worksheet Answers
 - Highlighting and Note-Taking Elements And Bonding Worksheet Answers
 - Interactive Elements Elements And Bonding Worksheet Answers
- 8. Staying Engaged with Elements And Bonding Worksheet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Elements And Bonding Worksheet Answers
- 9. Balancing eBooks and Physical Books Elements And Bonding Worksheet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Elements And Bonding Worksheet Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Elements And Bonding Worksheet Answers
 - Setting Reading Goals Elements And Bonding Worksheet Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Elements And Bonding Worksheet Answers

- Fact-Checking eBook Content of Elements And Bonding Worksheet Answers
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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