

Model 1 - White Light

	•	Color	Photon Energy (x 10 <sup>-21</sup> ) (J)	Wavelength Range (nm)	Speed (m/s)
Lightbulb (white light)	Prism	Reds	269-318	625-740	3.00 × 10°
		Oranges	318-337	590-625	3.00 × 10 <sup>8</sup>
	34 97	Yellows	337-352	565-590	3.00 × 10 <sup>4</sup>
( *	$\wedge$ /	Greens	352-382	520-565	$3.00 \times 10^{8}$
10/	$\checkmark \setminus \mathscr{U}$	Blues	382-452	440-520	$3.00 \times 10^{6}$
		Violets	452-523	380-440	$3.00 \times 10^{4}$

- 1. Trace the arrows in Model 1 and shade in the table with colored pencils where appropriate.
- 2. What happens to white light when it passes through a prism?

3. Why are the color labels in the table in Model 1 plural (i.e., "Reds" rather than "Red")?

4. Do all colors of light travel at the same speed?

5. Do all colors of light have the same energy? If no, which colors have the highest energy and the least energy, respectively?

warm colors have less energy cool colors have more

# **Electron Energy And Light Answer Key**

**Xiang Xie** 

## **Electron Energy And Light Answer Key:**

**Light & Sound (ENHANCED eBook)** Edward P. Ortleb,Richard Cadice,1993-09-01 This book presents a program of basic studies dealing with light and sound energy The sources and nature of light and sound are presented along with various characteristics of each phenomenon Topics include instruments that use and observe light and sound materials that affect light and sound and communication Each of the twelve teaching units in this book is introduced by a color transparency print books or PowerPoint slide eBooks that emphasizes the basic concept of the unit and presents questions for discussion Reproducible student pages provide reinforcement and follow up activities The teaching guide offers descriptions of the basic concepts to be presented background information suggestions for enrichment activities and a complete answer key

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