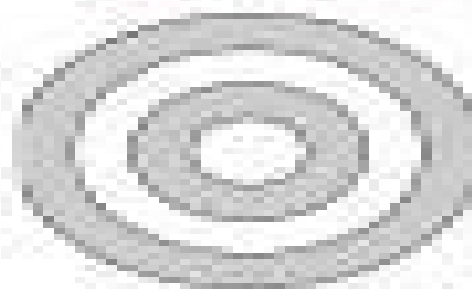


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1. Label the components of the diagram:



2. How many subatomic particles are present in the nucleus? _____
3. How many subatomic particles are shown in the electron shells? _____
4. Describe the subatomic particles: _____

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5. What is an orbital? _____
6. What probability is shown? _____
7. What probability is shown? _____
8. How many orbitals are shown? _____
9. Describe the orbitals shown: _____

10. What are the orbitals made of? _____

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11. What is the maximum configuration for each? _____
12. Describe the orbital for atoms and number of electrons: _____

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Ying Liu



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