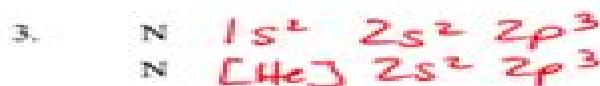


**8 • Electron Configurations & Periodicity****WRITING ELECTRON CONFIGURATIONS**

For each given element, fill in the orbital diagram and then write the electron configuration for the element.

1.	2.	3.	4.	5.	6.
Element: Ar # of e <sup>-</sup> 's: <u>18</u>	Element: Mg # of e <sup>-</sup> 's: <u>12</u>	Element: N # of e <sup>-</sup> 's: <u>7</u>	Element: Li # of e <sup>-</sup> 's: <u>3</u>	Element: P # of e <sup>-</sup> 's: <u>15</u>	Element: Cl # of e <sup>-</sup> 's: <u>17</u>

Write the electron configurations of each of these in long form and short form:



7. Fill in the orbital diagram for the element, Fe, and write the electron configuration of Fe in the long and short form.



# Electron Configuration And Periodicity Lab Answers

**Yicheng Fang**

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