

Gravimetric Analysis

Gravimetric analysis: a quantitative method for accurately determining the amount of a substance by selective precipitation of the substance from an aqueous solution.



Gravimetric analysis will be used in this lab to determine the identity of the cation in a sulfate salt in the following reaction:



The concept here is that if we add excess $BaCl_2$, the unknown sulfate becomes a limiting reactant which means all the sulfate in the unknown will be converted to the sulfate in the $BaSO_4$ precipitate. By using the concept of percent composition, we will be able to determine the identity of M using only two

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