

1. What volume of hydrogen gas at STP is produced from the reaction of 50.0g of Mg and 75.0 grams of HCl? How much of the excess reagent is left over (in grams)?



Do a standard mass to mass problem starting with each reactant

50.0 g Mg	$\frac{1 \text{ mol Mg}}{24.31 \text{ g Mg}}$	$\frac{1 \text{ mol H}_2}{1 \text{ mol Mg}}$	$\frac{22.4 \text{ L}}{1 \text{ mol H}_2}$	= 46.1 L H <sub>2</sub>
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75.0 g HCl	$\frac{1 \text{ mol HCl}}{36.46 \text{ g HCl}}$	$\frac{1 \text{ mole H}_2}{2 \text{ mol HCl}}$	$\frac{22.4 \text{ L}}{1 \text{ mol H}_2}$	= 23.0 L H <sub>2</sub>
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**HCl is the limiting reactant!!**

# Flinn Scientific Pogil Activity Limiting And Excess Reactants

**Lifelique**



**Flinn Scientific Pogil Activity Limiting And Excess Reactants:**

**Limiting Reactant and Percent** Lifeliqe,2019 This lesson plan covers analyzing chemical reactions in order to determine limiting reactants and excess reactants including calculating the amount of excess reactant calculate the theoretical yield of a reaction and calculating the percent yield of a reaction

## Whispering the Secrets of Language: An Emotional Quest through **Flinn Scientific Pogil Activity Limitting And Excess Reactants**

In a digitally-driven world where screens reign great and instant conversation drowns out the subtleties of language, the profound strategies and psychological subtleties hidden within words frequently go unheard. Yet, situated within the pages of **Flinn Scientific Pogil Activity Limitting And Excess Reactants** a fascinating fictional prize sporting with natural thoughts, lies an exceptional quest waiting to be undertaken. Penned by a talented wordsmith, this wonderful opus encourages readers on an introspective trip, gently unraveling the veiled truths and profound influence resonating within the very material of every word. Within the psychological depths with this emotional review, we can embark upon a sincere exploration of the book is primary themes, dissect their charming publishing model, and fail to the effective resonance it evokes deep within the recesses of readers hearts.

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