

DRAFT RELEASED FRQ ANSWERS AP CHEMISTRY 2014

(numbers in parentheses and red, represent *guessimate* points)

Question 1



(ii) Potassium and nitrate ions are in solution at the beginning of the reaction and at the end, do not change in any way (i.e., they are spectators), and as such can be omitted from the equation. (1)

(b) In order to ensure that the mass of the precipitate includes only the mass of the solid and none of the water, repeated drying will ensure that all of the water is removed before a final mass is recorded. (1)

(c) Less than. If the lead(II) nitrate were added in the stoichiometric ratio, the conc. of K^{+} and NO_3^{-} would be 1:1, but since the lead(II) nitrate is added in excess, the nitrate ions are present in the larger conc. (1)

(d) $1.698 - 1.462 = 0.236 \text{ g}$ of precipitate.

$$\text{Moles of precipitate} = \frac{0.236 \text{ g}}{(207.2 + 126.91 + 126.91 \text{ g mol}^{-1})} = \frac{0.236 \text{ g}}{461.02 \text{ g mol}^{-1}} = 5.12 \times 10^{-4} \text{ moles} \quad (1)$$

(e) Mass of I^{-} = (2) (moles of PbI_2) $(126.91 \text{ g mol}^{-1}) = 0.130 \text{ g}$

$$\text{Mass \% of I}^{-} = \frac{0.130 \text{ g}}{0.425 \text{ g}} \times 100 = \frac{30.6}{100} (100) = 30.6\% \quad (2)$$

(f) The same since the water plays no part in the mass of I^{-} present (this assumes all of the I^{-} is still precipitated by excess lead (II) nitrate, the tablets are the same and that the precipitate is dried completely as before etc.). (1)

(g) (i) Yes, since like lead(II) ions, silver ions also form an insoluble precipitate with iodide ions and since the K_{sp} is sufficiently small to suggest a precipitation will occur. (1)

(ii) No, IF after subtracting the mass of the dried precipitate + filter paper, from the mass of filter paper, yields a mass that has less than three significant figures, or

Yes, IF after subtracting the mass of the dried precipitate + filter paper, from the mass of filter paper, yields a mass that has at least three significant figures. (1)

(No is the expected answer I feel, BUT it does depend on the mass of the tablet IMO).

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1. Understanding the eBook Flinn Ap Chem 2014 Frq Answers
 - The Rise of Digital Reading Flinn Ap Chem 2014 Frq Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Flinn Ap Chem 2014 Frq Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Flinn Ap Chem 2014 Frq Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Flinn Ap Chem 2014 Frq Answers
 - Personalized Recommendations
 - Flinn Ap Chem 2014 Frq Answers User Reviews and Ratings
 - Flinn Ap Chem 2014 Frq Answers and Bestseller Lists
5. Accessing Flinn Ap Chem 2014 Frq Answers Free and Paid eBooks
 - Flinn Ap Chem 2014 Frq Answers Public Domain eBooks

- Flinn Ap Chem 2014 Frq Answers eBook Subscription Services
- Flinn Ap Chem 2014 Frq Answers Budget-Friendly Options
- 6. Navigating Flinn Ap Chem 2014 Frq Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Flinn Ap Chem 2014 Frq Answers Compatibility with Devices
 - Flinn Ap Chem 2014 Frq Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Flinn Ap Chem 2014 Frq Answers
 - Highlighting and Note-Taking Flinn Ap Chem 2014 Frq Answers
 - Interactive Elements Flinn Ap Chem 2014 Frq Answers
- 8. Staying Engaged with Flinn Ap Chem 2014 Frq Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Flinn Ap Chem 2014 Frq Answers
- 9. Balancing eBooks and Physical Books Flinn Ap Chem 2014 Frq Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Flinn Ap Chem 2014 Frq Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Flinn Ap Chem 2014 Frq Answers
 - Setting Reading Goals Flinn Ap Chem 2014 Frq Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Flinn Ap Chem 2014 Frq Answers
 - Fact-Checking eBook Content of Flinn Ap Chem 2014 Frq Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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