Lab 7: Flame Test Lab

The normal electron configuration of atoms or lons of an element is known as the "ground state". In this most stable energy state all of the electrons are in the lowest energy levels available. When atoms or ions in the ground state are heated to high temperatures some electrons may absorb energy and "jump" to a higher energy level. The element is then said to be in an "excited state". This excited configuration is unstable and the electrons will "fall back" to their normal positions of lower energy. As the electrons return to their normal levels the energy that was absorbed is now emitted in the form of electromagnetic radiation. Some of this radiation is in the form of visible light.

The color of this light can be used as a means of identifying the elements involved. This analysis is known as a flame test. Only metals, with their loosely held electrons, are excited by the flame of a bunsen burner. Thus flame tests are useful in the identification of metallic ions. Many metallic ions exhibit characteristic colors when vaporized by the burner flame. In this experiment characteristic colors of several different metallic ions will be observed and unknown ions will be identified by means of the flame test.

Before beginning the pre-lab questions below, be sure to read the procedure on the next page.

- Pre-Lab Questions:
 - 1. What is the difference between ground state and an excited state?

Grand state: all electrons are at the lawst possible Exitel state: Atoms/ Ing absorbenelar and 90 to

Hour:_

What does the word "emit" mean?

Enix 15 defines as being produced (i.e the en for the was assorbed is produced in the form of

EVERTHINGTON (3. In this experiment, where are the atoms getting their excess energy from?

> The electrons are gerting the excess energy from a bursen burner

4. Why do different atoms emit different colors of light?

Atoms emit different was because of their Chambral makeup

Why is it necessary to clean the nichrome wires between each flame test?

It is necessary to clean the wires between the flame tests because if you don't, the atoms might not display their characterists color because they were mixed with

Flame Test Lab Answers

Brian F. Woodfield, Matthew C. Asplunc

Flame Test Lab Answers:

The Science I Know Suzanna Roman-Oliver, 2024-07-08 The Science I Know Culturally Relevant Science Lessons from Secondary Classrooms is a collection of culturally relevant lesson plans written by secondary science teachers Each lesson discusses how the tenets of academic success cultural competence and critical consciousness that are part of the theory of Culturally Relevant Pedagogy CRP are addressed Ladson Billings 1995 Additionally each lesson plan is structured following the 5E learning cycle Bybee 2006 and aligned to the Next Generation Science Standards NAS 2012 The goal of this book is to help science teachers understand how to go about designing lessons that are culturally relevant The hope is that the lessons that are detailed in each chapter will inspire teachers to draw the cultural knowledge from their students and capitalize on it when designing science lessons After an introductory chapter that discusses how science education has shifted in recent decades to address the needs of diverse students the main body of the text is divided into three sections. The first part introduces Culturally Relevant Pedagogy CRP as a framework this is important for those readers unfamiliar with Gloria Ladson Billings work It addresses and discusses the three tenets of CRP Academic Success Cultural Competence and Critical Consciousness and it includes an explanation of how each area can be observed and addressed in science education specifically The second part features lesson plans from secondary science classrooms written by teachers from different subject areas i e life science physical science earth science etc The lesson plans follow the 5E Instructional Model Bybee et al 2006 This model promotes inquiry by guiding teachers in the design of lesson plans that are based upon cognitive psychology constructivist learning theory and best practices in science teaching Duran Education and Culture Learner Differences Secondary Science Pedagogy Culturally Relevant Science and Multicultural Education Chemistry in the Laboratory James M. Postma, Julian L. Robert, J. Leland Hollenberg, 2004-03-12 This clearly written class tested manual has long given students hands on experience covering all the essential topics in general chemistry Stand alone experiments provide all the background introduction necessary to work with any general chemistry text This revised edition offers new experiments and expanded information on applications to real world situations Lab Experiments in Introductory Chemistry Phil Reedy, Donald J. Wink, Sharon Fetzer-Gislason, 2003-03-21 The manual contains laboratory experiments written specifically for the prep chem lab as well as for the general chemistry course Available as a complete manual or custom published athttp **Laboratory Manual for Principles of General Chemistry** Jo Allan Beran, 2010-11-01 This custompub whfreeman com new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures **PE Lab Exp(Noncons)Mod Chem 90** Tzimopoulo,1990 Holt Chemistry Ralph Thomas Myers, 2004 The Life Cycle of Everyday Stuff Mike Reeske, Shirley Watt Ireton, National

Science Teachers Association,2001 In the context of life cycles these units use central science concepts to explore the energy raw materials and waste issues that are the history of any manufactured product As students consider the trade offs made at each step they will learn to recognize the decisions made to balance economic developmental and environmental needs

<u>ChemDiscovery Teacher Edition</u> Olga I. Agapova,2002 **Popular Science**,1934-03 Popular Science gives our readers the information and tools to improve their technology and their world The core belief that Popular Science and our readers share The future is going to be better and science and technology are the driving forces that will help make it better

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