

Name : _____ Date : _____

Score : _____



Electron Configuration



1. Determine what elements are denoted by the following configurations.

i. $1s^2 2s^2 2p^6 3s^2 3p^4$ _____

ii. $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^5$ _____

iii. $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6$ _____

iv. $[\text{Kr}] 5s^2 4d^{10} 5p^3$ _____

v. $[\text{Ne}] 3s^2 3p^1$ _____

2. Write the full electron configuration (ex., $1s^2 2s^1$) of the following elements?

i. Nitrogen _____

ii. Silicon _____

iii. Calcium _____

iv. Strontium _____

v. Bismuth _____

3. Write the abbreviated electron configuration (ex., $[\text{He}] 2s^1$) of the following element?

i. Fluorine _____

ii. Magnesium _____

iii. Selenium _____

iv. Rubidium _____

v. Lead _____

4. If each orbital can hold a maximum of two electrons, how many can each of the following hold?

i. $2s$ _____ ii. $5p$ _____ iii. $4f$ _____ iv. $3d$ _____ v. $4d$ _____

5. What is the shape of an s orbital? _____

6. What is the shape of a p orbital? _____

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Configuration - What You Need to Know Dale Rich, 2014-10-15 It's a brand new Electron configuration world There has never been a Electron configuration Guide like this It contains 148 answers much more than you can imagine comprehensive answers and extensive details and references with insights that have never before been offered in print Get the information you need fast This all embracing guide offers a thorough view of key knowledge and detailed insight This Guide introduces what you want to know about Electron configuration A quick look inside of some of the subjects covered Extended periodic table 8s elements List of data references for chemical elements References for chemical elements Ferrocene Structure and bonding Coordination complex Geometry History of electromagnetic theory Lorentz and Poincare Ion Formation of polyatomic and molecular ions Valence electrons Copernicium Physical and atomic Cadmium Mol Phys Term symbol Valence electron Electron configuration Pnictogen Chemical Electronic configuration Electron configuration in molecules Group 11 elements Hund's rules Octet rule Explanation in quantum theory Group 11 element Characteristics Hassium Physical and atomic Alkali metal Extended periodic table Superactinides Neutron absorption Neutron absorbers Flerovium Nuclear stability and isotopes Lewis acid Comparison with Bronsted Lowry Theory Ununoctium Calculated atomic and physical properties Tunnel diode f block Isotopes of strontium Ytterbium Aufbau principle Atoms Energy levels Pauli exclusion principle Atoms and the Pauli principle Electron shell List of elements with electrons per shell Periodic table Layout Rutherfordium Chemical Antimony Group 8 element Lanthanum Chemical properties Extended periodic table Eka superactinides Manganese Promethium Physical properties and much more **Facilitating Conceptual Change in**

Students' Understanding of the Periodic Table Mansoor Niaz, Marniev Luiggi, 2013-07-16 This book is about how students are taught the periodic table. It reviews aspects of the periodic table's development using the history and philosophy of science. The teaching method presented in this book is ideal for teaching the subject in high school and at introductory university level. Chemistry students taught in this new experimental way are compared with those taught in the traditional way and the author describes how tests found more conceptual responses from the experimental group than the control group. The historical aspects of importance to this teaching method are the role of the Karlsruhe Congress of 1860, the accommodation of the chemical elements in the periodic table, prediction of elements that were discovered later, corrections of atomic weights, periodicity in the periodic table as a function of the atomic theory and the accommodation of argon. The experimental group of students participated in various activities including discussion of various aspects related to the history and philosophy of science, construction of concept maps and their evaluation by the students, PowerPoint presentations and interviews with volunteer students.

Reflections on the Periodic Table Jaime Aguirre, 2003 Symphony of matter is a PowerPoint presentation featuring the construction of the periodic table based on the ground electronic configurations of 118 chemical elements. p 32

Electrons in Matter and Link to Redox Chemistry Malika Ammam, 2018-03-05 As a teacher of physical chemistry, I noticed that students even in advanced classes have difficulties in understanding the basics of redox chemistry. In this Section 2, I attempted to summarize some fundamental principles related to electrons including the atomic structure, periodic table, main electronic models, electronic configuration and the energy levels of electrons. The identification of the electronic configurations of elements and visualization of electrons in their outer shells are very important concepts in understanding redox processes. To further clarify the discussed concepts, numerous questions and problems with detailed answers are provided. Most of these questions are formulated by students like you. I believe that this Section 2 would greatly help students with levels varying from high school to advanced university classes.

Electronic Structure, Properties, and the Periodic Law Harry Hall Sisler, 1963

The Periodic Table in Minutes Dan Green, 2016-09-06 An icon of science, the Periodic Table defines the fundamental chemistry of everything in the universe. In this compact yet comprehensive guide, Dan Green outlines the history, development and workings of the table, shows how its design reflects and illuminates the organisation of all matter and even explains what it has to tell us about the chemistry of distant stars and of our own bodies. Contents include an individual entry for every known element detailing properties, uses and key data, and sections on the patterns and groups of the famous table as well as explanations of basic chemistry concepts such as elements and compounds, atomic structure, chemical bonds, reactions and radioactivity amongst many others.

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