

LABORATORY EVALUATION OF STRENGTHENING TECHNIQUES — *John J. Lesko*

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- 1.1. What are the two main categories of chemical reactions? (10 marks)
 1.2. Explain the term "stoichiometry" and its importance in chemistry. (10 marks)
 1.3. Write the balanced chemical equation for the reaction between hydrogen and oxygen to form water. (10 marks)
 1.4. Define the term "molar mass" and calculate the molar mass of water (H₂O). (10 marks)
 1.5. What is the difference between a chemical reaction and a physical change? (10 marks)
 1.6. Explain the concept of "limiting reagent" in a chemical reaction. (10 marks)
 1.7. Calculate the mass of water produced when 2 moles of hydrogen gas react with 1 mole of oxygen gas. (10 marks)
 1.8. Write the balanced chemical equation for the combustion of methane (CH₄). (10 marks)
 1.9. Define the term "pH" and explain its significance in chemistry. (10 marks)
 1.10. Calculate the pH of a 0.1 M solution of hydrochloric acid (HCl). (10 marks)

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Mike Smith



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