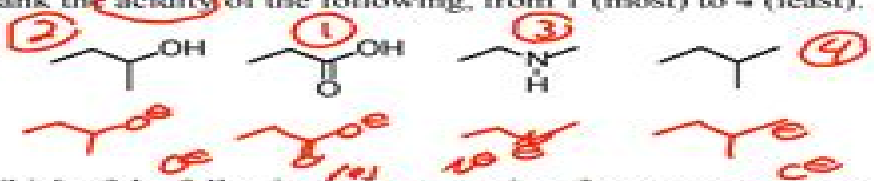


6. Rank the acidity of the following, from 1 (most) to 4 (least). (4pt)

Think Anion Base

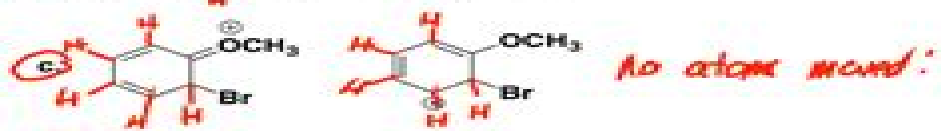


3 Acidity Factors:  
 1. charge (not relevant here)  
 2. E<sub>neg</sub>  
 3. Resonance

7. Which of the following represent pairs of resonance structures? (4pt)



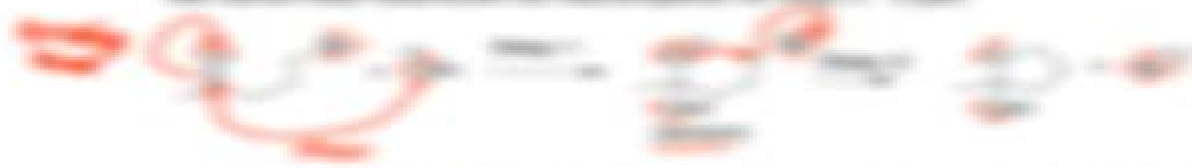
No atoms can move!



→ d. Both a and c

e. a, b, and c are all resonance structures.

8. Draw the Lewis structure for the following molecule, showing all lone pairs and formal charges. (4pt)



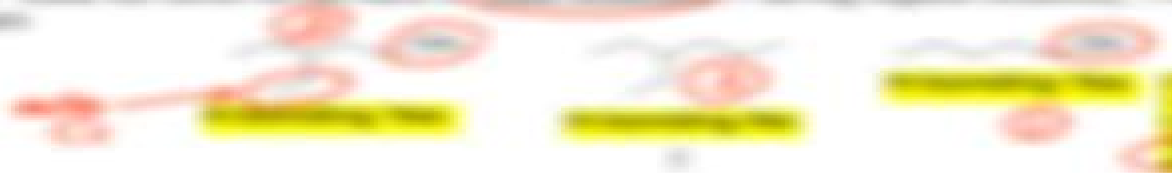
Lewis structure  
 Lone pairs  
 Formal charges

9. Draw the Lewis structure for the following molecule, showing all lone pairs and formal charges. (4pt)



Lewis structure  
 Lone pairs  
 Formal charges

10. Draw the Lewis structure for the following molecule, showing all lone pairs and formal charges. (4pt)



Lewis structure  
 Lone pairs  
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# Klein Organic Chemistry Answer Key

**Léo Pomar**



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