

Prelim Topic 1: The Chemical Earth

1. The living and non-living components of the Earth contain mixtures

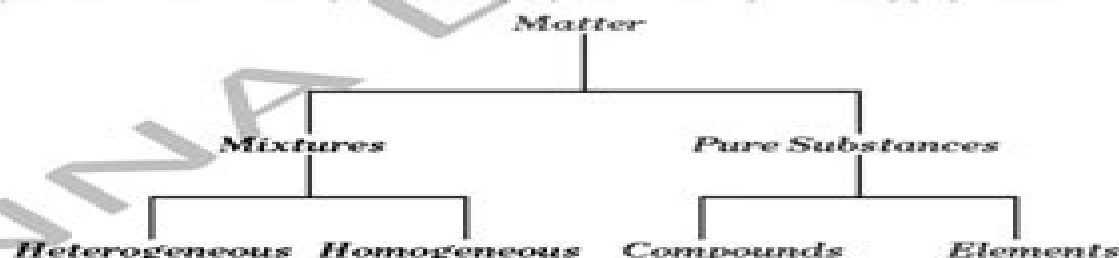
- ❖ Identify the difference between elements, compounds and mixtures in terms of particle theory.

Particle theory

All matter are made up of tiny particles which are continually moving.

Types of matter

1. Pure substance contains fixed compositions, fixed properties and cannot be decomposed by simple physical separation techniques.
 - a. **Elements** are composed of atoms or molecules
 - Simplest pure substance consisting of one type of atom
 - Cannot decompose
 - Exists as **molecules** (substances with two or more atoms of the same type chemically combined)
 - b. **Compounds** are composed of fixed numbers of atoms of different elements
 - Composed of two or more elements chemically bonded together
 - Fixed number of atoms of each component element
 - Decompose into component elements through chemical means
2. Mixture (Impure) have various particle types and compositions.
 - a. **Homogeneous mixtures** have uniform composition and properties throughout eg. Air, water
 - b. **Heterogeneous mixtures** have variable composition and properties throughout eg. Blood, concrete
 - Components of mixture keep their own properties and separated by physical means eg. (filtering)



- ❖ Identify that the biosphere, lithosphere, hydrosphere and atmosphere contain examples of mixtures of elements and compounds

| Biosphere | | Major form | Common elements |
|-----------|---|---------------------------|---|
| | - the region on Earth where living organisms are found | Water | Hydrogen, oxygen |
| | -most living things are composed of cells which contain complex carbon compounds such as carbohydrates, fats and proteins | Calcium phosphate (bones) | Phosphorus, calcium |
| | | Carbohydrates, fats | Carbon, hydrogen, oxygen |
| | | Proteins | Carbon, hydrogen, oxygen, nitrogen, sulphur |

Kiss Prelim Chemistry Topic Chemical Earth Answers

M Woodhall



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