

Religion

Please match the following religion with the correct definition

1. Buddhism **b**
2. Christianity **a**
3. Confucius **c**

4. Islam **c**
5. Judaism **e**
6. Confucianism **d**

- ✓ Based on the person and teachings of Jesus, the Bible is the holy book
- ✓ Judaism/other religions is the founding, membership, religious that encompasses a variety of traditions
- ✓ Mohammed is the founder, respects the holy book, Quran, believes in heaven by following the 5 pillars

- ✓ Confucius is the founding father, the moral order and good government should be based on strong family relationships
- ✓ Abraham is the founding father is the holy book, developed among the ancient traditions
- ✓ This is more study, emphasizing harmony from the material world through purification

Please answer the following questions

Where do prehistoric people believe is the birthplace of humankind? (where?)

- **Great Rift Valley in Africa** - **around 100,000 years ago**

- **the beginning of farming** - **lead to specialization of jobs**
The use of technology to work a new land (farming) and the introduction of farming.

- **Stone tools**

around the 100,000 years ago

- **Around 3000 BC**

around the 100,000 years ago

- **near rivers/river valleys** - **stable source of water**

around the 100,000 years ago

- **Egypt - Nile River**

- **Mesopotamia - Tigris & Euphrates River** - **Indus River Valley - Indus River**

around the 100,000 years ago

- **People live in cities** - **complex social institutions** - **writing** - **technology**

around the 100,000 years ago

- **Government & Religion began**

around the 100,000 years ago

- **Monarchy / theocracy**

- **Polytheistic** - **Zoroastrianism**

around the 100,000 years ago

- **Zigurat**

- **Cuneiform** - **Phoenician Alphabet** - **Hammurabi's Code** - **Wheels**

around the 100,000 years ago

- **Hammurabi's Law Code**

around the 100,000 years ago

- **on Nile River**

around the 100,000 years ago

- **Pharaoh with absolute rule** - **Pharaoh**

- **Pharaoh had all the power**

- **Polytheistic**

around the 100,000 years ago

- **Pyramids, hieroglyphics, pharaohs, medicine, astronomy**

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Table of Contents Eng Odysseyware Semeser Exam Answers

1. Understanding the eBook Eng Odysseyware Semeser Exam Answers
 - The Rise of Digital Reading Eng Odysseyware Semeser Exam Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Eng Odysseyware Semeser Exam Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Eng Odysseyware Semeser Exam Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Eng Odysseyware Semeser Exam Answers
 - Personalized Recommendations
 - Eng Odysseyware Semeser Exam Answers User Reviews and Ratings
 - Eng Odysseyware Semeser Exam Answers and Bestseller Lists
5. Accessing Eng Odysseyware Semeser Exam Answers Free and Paid eBooks
 - Eng Odysseyware Semeser Exam Answers Public Domain eBooks
 - Eng Odysseyware Semeser Exam Answers eBook Subscription Services
 - Eng Odysseyware Semeser Exam Answers Budget-Friendly Options
6. Navigating Eng Odysseyware Semeser Exam Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Eng Odysseyware Semeser Exam Answers Compatibility with Devices
 - Eng Odysseyware Semeser Exam Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Eng Odysseyware Semeser Exam Answers
 - Highlighting and Note-Taking Eng Odysseyware Semeser Exam Answers
 - Interactive Elements Eng Odysseyware Semeser Exam Answers
8. Staying Engaged with Eng Odysseyware Semeser Exam Answers

- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Eng Odysseyware Semeser Exam Answers
9. Balancing eBooks and Physical Books Eng Odysseyware Semeser Exam Answers
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Eng Odysseyware Semeser Exam Answers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Eng Odysseyware Semeser Exam Answers
- Setting Reading Goals Eng Odysseyware Semeser Exam Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Eng Odysseyware Semeser Exam Answers
- Fact-Checking eBook Content of Eng Odysseyware Semeser Exam Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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for the nitration 4 01 ml of nitric acid 4 00 ml of sulfuric acid and 4 56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4 56 ml of bromobenzene was used the amount of product predicted to be recovered was 8 7533 g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction eas particularly nitration of methyl benzoate go give methyl m nitrobenzoate thus determine the weight melting point and percentage yield of the pure methyl m nitrobenzoate abstract

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its

reactions will be slower than the identical reactions of benzene

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section 20711 lab experiment nitration of bromobenzene introduction the chemical reaction of nitration consists of a nitro group being added to or substituted in a molecule nitration can basically be carried out by a mixture of concentrated nitric acid and sulfuric acid this mixture is useful to obtain the active nitronium ion

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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nitration of bromobenzene lab report improved essays 731 words 3 pages open document essay sample check writing quality show more this experiment was performed in order to demonstrate the principles of nitration through electrophilic aromatic substitution it also examined the effects of bromine on an aromatic ring ortho para vs meso directing

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then actual is your actual findings yield actual theoretical $\times 100$ what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area theory and explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 °C

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1-bromo-4-nitrobenzene

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study with quizlet and memorize flashcards containing terms like reacting sulfuric and nitric acid together produces the nitronium ion alone nitric acid is not a strong enough electrophile to react with bromobenzene sulfuric acid is added to the nitric acid the reverse rxn can overheat and boil easily includes a claisen adaptor and

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture hno 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0 6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accomodate their electrons demonstration of creating the p arenium cation by to nitration of bromobenzene

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nitrobromobenzene recrystallization was used to purify 4 nitrobromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion no2 which acts as the electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4 0ml of nitric acid and 4 0ml of sulfuric acid and will then react with 4 5ml of bromobenzene this will produce o bromonitrobenzene and p bromonitrobenzene

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