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## Chemistry Comes Alive

### Objectives

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#### Part 1: Basic Chemistry

##### Definition of Concepts: Matter and Energy

1. Differentiate between matter and energy and between potential energy and kinetic energy.
2. Describe the major energy forms.

##### Composition of Matter: Atoms and Elements

3. Define chemical element and list the four elements that form the bulk of body matter.
4. Define atom. List the subatomic particles, and describe their relative masses, charges, and positions in the atom.
5. Define atomic number, atomic mass, atomic weight, isotope, and radioisotope.

##### How Matter Is Combined: Molecules and Mixtures

6. Define molecule, and distinguish between a compound and a mixture.
7. Compare solutions, colloids, and suspensions.

##### Chemical Bonds

8. Explain the role of electrons in chemical bonding and in relation to the octet rule.
9. Differentiate among ionic, covalent, and hydrogen bonds.
10. Compare and contrast polar and nonpolar compounds.

##### Chemical Reactions

11. Define the three major types of chemical reactions: synthesis, decomposition, and exchange. Comment on the nature of oxidation-reduction reactions and their importance.
12. Explain why chemical reactions in the body are often irreversible.

# Marieb Lab Manual 10th Edition Answer Key

**Darrel Hess, Tom Lee**  
**McKnight, Stephen J. Stadler**



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