

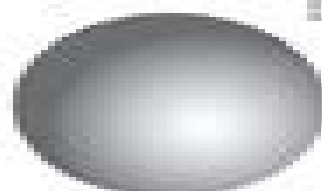
Name: \_\_\_\_\_

Class: \_\_\_\_\_

Date: \_\_\_\_\_

Identify which atomic model is shown in each picture and write the name below the picture. Write the scientist responsible for the model below each picture as well. There is a word bank below to help you.

## History of the Atom

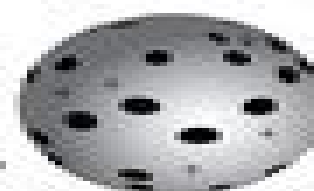


Indivisible atom

1a. Model: \_\_\_\_\_

b. Scientist(s) responsible: \_\_\_\_\_

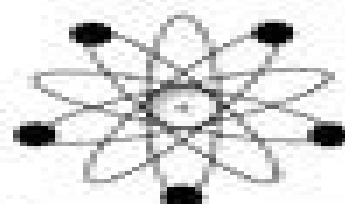
Scattered negatively charged small particles within the body



positively charged low density 'soup'

2a. Model: \_\_\_\_\_

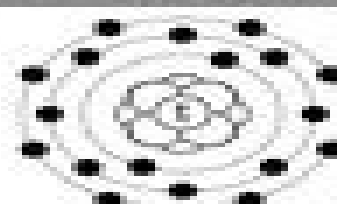
b. Scientist(s) responsible: \_\_\_\_\_



positively charged center with orbiting negatively charged particles

3a. Model: \_\_\_\_\_

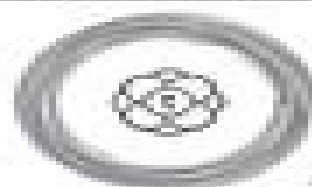
b. Scientist(s) responsible: \_\_\_\_\_



Nucleus with protons, neutrons, and electrons must orbit at defined energy levels and electrons away from the nucleus

4a. Model: \_\_\_\_\_

b. Scientist(s) responsible: \_\_\_\_\_



Nucleus with electrons that have calculated probable areas where they are located, but no exact location can be determined

5a. Model: \_\_\_\_\_

b. Scientist(s) responsible: \_\_\_\_\_

6. Now examine the following statement's from Dalton's Atomic Theory. Circle true or false for each statement and if it's false, write below what you know about the atom today that contradicts his theory

A. All matter is made of atoms.  
True or False

B. Atoms are indivisible and indestructible.  
True or False

C. All atoms of a given element are identical in mass and properties.  
True or False

D. Compounds are formed by a combination of two or more different kinds of atoms.  
True or False

Models that is chronological order 2  
 primitive model  
 simple, first atomic theory  
 plum pudding model  
 Bohr model  
 Electron Cloud model

Scientists that is order 2  
 Ernest Rutherford  
 J.J. Thomson  
 John Dalton  
 Modern Scientists  
 Niels Bohr

# History Of Atom Worksheet Answers

**L Darling-Hammond**



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*Science Spectrum* Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2003-03 *Radiologic Science* Stewart C. Bushong, 1984 **The Science Teacher**, 2001 **Pageant World History** Gerald Leinwand, 1990 Dear students I want to share a dream with you I dreamed that a young person of 14 whom I was going to be teaching would become president of the United States during the first half of the twenty first century As a teacher I was struck by this immense responsibility What should I teach my student about the world as preparation for this awesome task How could my world history class help this person to mature into an intelligent and humane president and leader of the free world

**Glencoe Science** McGraw-Hill Staff, 2001-08 **The American Biology Teacher**, 2003 *Life and Liberty* Philip Roden, 1987 **New Scientist**, 1993 **New Scientist and Science Journal**, 1993-08 *Data Sources*, 2000 **Library Media Connection**, 1995 **Whitaker's Five-year Cumulative Book List**, 1968 **From Atomos to Atom** Andreas Gerardus Maria van Melsen, 1960 *Atoms*

Timothy L. Biel,1990 Explores the history and development of atomic theory from Democritus to Einstein and the present presenting physics and chemistry experiments done by famous scientists and discussing nuclear power fusion and fission

The Progress of Atomic Theory Albert Cushing Crehore,1926

*The Question of the Atom* Mary J. Nye,1983-01-01

**Atomic and Molecular Structure** Joel Chaffee,2011-01-15 Learn about the atom what it is the people responsible for helping us understand it and how it affects us in the world today

A New View of the Origin of Dalton's Atomic Theory

Roscoe,1896

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