

Answer: D. 5.0

7. Calculate the concentration of hydrogen ions in the following acetic acid solution. $pK_a = 4.76$

$$[CH_3COOH] / [CH_3COO^-] = 0.5$$

A. $7.89 \times 10^{-5} M$

B. $6.31 \times 10^{-5} M$

C. $8.39 \times 10^{-5} M$

D. $1.26 \times 10^{-5} M$

Answer: C. $8.39 \times 10^{-5} M$

8. Which group contains all of the following types of bonds (except)

A. $\sigma(C_{sp^2}-C_{sp^2})$

B. $\sigma(C_{sp^2}-C_{sp})$

C. $\sigma(C_{sp^2}-C_{sp^3})$

D. $\sigma(C_{sp^2}-C_{sp^2})$

E. $\sigma(C_{sp^2}-C_{sp^2})$

Answer: E. $\sigma(C_{sp^2}-C_{sp^2})$

9. The degree of unsaturation for cyclopentadiene is



A. 0

B. 2

C. 4

D. 1

E. 3

Answer: C. 4

Gre Chemistry Guide

Michael Hoelscher



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