

HYDROLYSIS OF SALTS AND pH OF BUFFER SOLUTIONS

INTRODUCTION

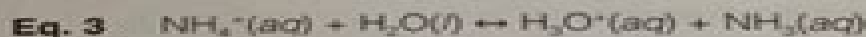
The purpose of this experiment is to determine the pH of various concentrations of acids, bases, salts, and buffer solutions. The data will be used to determine the acid-base nature in the hydrolysis of several salts and behavior of buffer solutions. Recall that salts are the product of neutralization reactions between acids and bases, generically shown below with an acid, HA, and base, B.



As nearly all salts are strong electrolytes, they exist in solution as ions. Many of these ions then can react with water to produce an acidic or basic solution. This reaction of salts with water is called a hydrolysis reaction. An anion, such as the generic A^- , is the conjugate base of the weak acid HA. These anions (A^-) will then react with water to generate OH^- .



Cations such as NH_4^+ or highly charged metals like Fe^{3+} come from weak bases and react with water to form H^+ (H_3O^+ , hydronium) ions.



HYDROLYSIS OF ANIONS: BASIC SOLUTIONS

First, let us focus on the hydrolysis of anions. For example, the nitrite ion (NO_2^-), which will react with water (like a weak base, for example NH_3) to generate hydroxide and thus a basic solution (pH greater than 7):



Hydrolysis Of Salts Lab Report Answers

Ian Pickup



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This book delves into Hydrolysis Of Salts Lab Report Answers. Hydrolysis Of Salts Lab Report Answers is a crucial topic that must be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Hydrolysis Of Salts Lab Report Answers, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Hydrolysis Of Salts Lab Report Answers
- Chapter 2: Essential Elements of Hydrolysis Of Salts Lab Report Answers
- Chapter 3: Hydrolysis Of Salts Lab Report Answers in Everyday Life
- Chapter 4: Hydrolysis Of Salts Lab Report Answers in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Hydrolysis Of Salts Lab Report Answers. The first chapter will explore what Hydrolysis Of Salts Lab Report Answers is, why Hydrolysis Of Salts Lab Report Answers is vital, and how to effectively learn about Hydrolysis Of Salts Lab Report Answers.
3. In chapter 2, the author will delve into the foundational concepts of Hydrolysis Of Salts Lab Report Answers. This chapter will elucidate the essential principles that need to be understood to grasp Hydrolysis Of Salts Lab Report Answers in its entirety.
4. In chapter 3, the author will examine the practical applications of Hydrolysis Of Salts Lab Report Answers in daily life. The third chapter will showcase real-world examples of how Hydrolysis Of Salts Lab Report Answers can be effectively utilized in everyday scenarios.
5. In chapter 4, the author will scrutinize the relevance of Hydrolysis Of Salts Lab Report Answers in specific contexts. This chapter will explore how Hydrolysis Of Salts Lab Report Answers is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, this book will draw a conclusion about Hydrolysis Of Salts Lab Report Answers. This chapter will summarize the key points that have been discussed throughout the book.
- The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Hydrolysis Of Salts Lab Report Answers.

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