

**Assignment #2 – Compound Names and Formulas**  
**Multi-valent ions only**

A. Write the correct formula for the following compounds, all of which have been named using the modern Roman Numeral Method. The ionic charge is given after the first element (metallic) in Roman Numerals.

1. copper (II) oxide CuO
2. mercury (I) oxide Hg<sub>2</sub>O
3. gold (III) chloride AuCl<sub>3</sub>
4. thallium (III) bromide TlBr<sub>3</sub>
5. bismuth (V) oxide Bi<sub>2</sub>O<sub>5</sub>
6. terbium (III) oxide Tb<sub>2</sub>O<sub>3</sub>
7. uranium (VI) oxide UO<sub>3</sub>
8. protactinium (IV) bromide PaBr<sub>4</sub>
9. cerium (III) oxide Ce<sub>2</sub>O<sub>3</sub>
10. arsenic (V) sulphide As<sub>2</sub>S<sub>5</sub>

11. manganese (III) oxide Mn<sub>2</sub>O<sub>3</sub>
12. vanadium (II) bromide VBr<sub>2</sub>
13. niobium(V) oxide Nb<sub>2</sub>O<sub>5</sub>
14. titanium (III) oxide Ti<sub>2</sub>O<sub>3</sub>
15. titanium (III) nitride TiN
16. iron (II) oxide FeO
17. cobalt (II) phosphide Co<sub>3</sub>P<sub>2</sub>
18. tin (II) oxide SnO
19. thulium (II) bromide TmBr<sub>2</sub>
20. copper (II) bromide CuBr<sub>2</sub>

B. Name the compound using the Roman Numeral Method and balance the ionic charges.

1. SnCl<sub>4</sub> Tin (IV) Chloride
2. BiBr<sub>5</sub> Bismuth (V) Bromide
3. PoO<sub>2</sub> Polonium (IV) Oxide
4. PbI<sub>2</sub> Lead (II) Iodide
5. HgO Mercury (II) Oxide
6. HgCl Mercury (I) Chloride
7. Au<sub>2</sub>O<sub>3</sub> Gold (III) Oxide
8. FeCl<sub>2</sub> Iron (II) Chloride

9. PdF<sub>4</sub> Palladium (IV) Fluoride
10. Os<sub>2</sub>O<sub>3</sub> Osmium (III) Oxide
11. MoBr<sub>2</sub> Molybdenum (II) Bromide
12. VCl<sub>5</sub> Vanadium (V) Chloride
13. Mn<sub>2</sub>O<sub>3</sub> Manganese (III) Oxide
14. CoO Cobalt (II) Oxide
15. Np<sub>2</sub>O<sub>3</sub> Neptunium (III) Oxide
16. V<sub>2</sub>O<sub>5</sub> Vanadium (V) Oxide

# Naming Ionic Compounds Practice Answer Key

**D Kirk**



## **Naming Ionic Compounds Practice Answer Key:**

## Unveiling the Magic of Words: A Review of "**Naming Ionic Compounds Practice Answer Key**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their capability to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Naming Ionic Compounds Practice Answer Key**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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### **Table of Contents Naming Ionic Compounds Practice Answer Key**

1. Understanding the eBook Naming Ionic Compounds Practice Answer Key
  - The Rise of Digital Reading Naming Ionic Compounds Practice Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Naming Ionic Compounds Practice Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Naming Ionic Compounds Practice Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Naming Ionic Compounds Practice Answer Key
  - Personalized Recommendations
  - Naming Ionic Compounds Practice Answer Key User Reviews and Ratings
  - Naming Ionic Compounds Practice Answer Key and Bestseller Lists

5. Accessing Naming Ionic Compounds Practice Answer Key Free and Paid eBooks
  - Naming Ionic Compounds Practice Answer Key Public Domain eBooks
  - Naming Ionic Compounds Practice Answer Key eBook Subscription Services
  - Naming Ionic Compounds Practice Answer Key Budget-Friendly Options
6. Navigating Naming Ionic Compounds Practice Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Naming Ionic Compounds Practice Answer Key Compatibility with Devices
  - Naming Ionic Compounds Practice Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Naming Ionic Compounds Practice Answer Key
  - Highlighting and Note-Taking Naming Ionic Compounds Practice Answer Key
  - Interactive Elements Naming Ionic Compounds Practice Answer Key
8. Staying Engaged with Naming Ionic Compounds Practice Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Naming Ionic Compounds Practice Answer Key
9. Balancing eBooks and Physical Books Naming Ionic Compounds Practice Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Naming Ionic Compounds Practice Answer Key
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Naming Ionic Compounds Practice Answer Key
  - Setting Reading Goals Naming Ionic Compounds Practice Answer Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Naming Ionic Compounds Practice Answer Key
  - Fact-Checking eBook Content of Naming Ionic Compounds Practice Answer Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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