

Group: **Chiller**Part Number: **331375401**Date: **July 2007**Supersedes: **OM AGSD-1**

Air-Cooled Screw Compressor Chiller

AGS 225D through AGS 450D**Software Version AGSG30101D****60 Hertz, R-134a**

Mcquay Screw Compressor Ags Service Manual

Brendan G. Carr



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web the molar volume will be the volume occupied by one molar mass 32 g of the liquid expressing the density in liters instead of ml we have $v_m = \frac{32 \text{ g mol}^{-1}}{1790 \text{ g l}^{-1}} = 0.0405 \text{ l mol}^{-1}$

web the molar volume of a gas from avogadro s law the volume is directly proportional to the number of moles the important point to keep in mind is that n can be moles of any gas since one mole always contain 6.02×10^{23} molecules and one molecule of any gas at the same temperature will have the same kinetic energy based on this principle it is found

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web the molar volume has the si unit of cubic metres per mole $\text{m}^3 \text{mol}^{-1}$ although it is more typical to use the units cubic decimetres per mole $\text{dm}^3 \text{mol}^{-1}$ for gases and cubic centimetres per mole $\text{cm}^3 \text{mol}^{-1}$ for liquids and solids

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