

The fundamental ideas in chemistry

1.	electron nucleus neutron	each for 1 mark		
				[3]
2.	(a) (i) B		1	
	(ii) D		1	
	(b) A and B – only one type of atom		1	
	C and D – more than one type of atom		1	
	accept element for atom			
	ignore the word 'mixture'			
	(chemically) bonded		1	
	accept (chemically) joined or similar idea of joined			
				[5]
3.	B carbon monoxide		1	
	CO	accept carbon oxide do not credit carbon dioxide do not credit if any superscripts or subscripts used but accept C ₂ O ₂ accept OC do not credit if obviously lower case	1	
	C water		1	
	H ₂ O	accept hydrogen oxide do not accept hydrogen hydroxide do not credit if obviously lower case or if 2 not subscript do not accept HOH accept OH ₂	1	
	D ammonia	do not accept ammonium	1	
	NH ₃	do not credit if obviously lower case, or if 3 not subscript accept nitrogen hydride or hydrogen nitride do not accept hydrogen nitrate or nitrite allow H ₃ N	1	
				[6]
4.	• correct reactants (i.e. sodium + water either way round)			
	• correct products (i.e. sodium hydroxide + hydrogen, either way round)			
	• arrow → / =	do not allow produce/makes or similar do not allow symbols or formulae		
		each for 1 mark		
				[3]
5.	correct use of 'react'/reaction/reactants/combine (not mixed/added/join)		3	
	correct use of 'produce'/products/gives/forms/makes/creates reactants			
	correctly identified			
	each for 1 mark			
	products correctly identified		1	

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Sabine Zange



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