

Skills Worksheet

Concept Review Chapter 5: Section 1

Section: Simple Ions

Complete each statement below by choosing a term from the following list. Terms may be used more than once.

10

11

ion

octet

12

13

anion

cation

1. An atom or group of atoms that has a positive or negative electric charge because it has lost or gained electrons is a(n) ion.
2. The sodium ion has 11 protons, 12 neutrons, and 10 electrons.
3. In most chemical reactions, atoms tend to match the outer electron configuration of the noble gases. This is called the octet rule.

Complete each statement below by writing the correct term or phrase.

4. An anion is an ion with a negative charge.
5. Valence electrons occupy the outermost energy level of an atom.
6. The chloride ion and the chlorine atom have different electron configurations.
7. A cation is an ion with a positive charge.
8. Many stable ions have an electron configuration of a noble gas.

Answer the following questions in the space provided.

9. How do the outer-shell electron configurations for ions of Group 1, Group 2, and Group 15, Group 16, and Group 17 elements compare with those of the noble gases?

They are the same. Group 1 + 2 ions have noble gas configuration of previous row. Group 15, 16 + 17 ions have noble gas configuration in their row.

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Holt Chemistry Ralph Thomas Myers, 2004 *Modern Chemistry* Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 **The School Science Review**, 2003 **Electropharmacology** Hendrik Keyzer, George M. Eckert, Felix Gutmann, 1990-09-25 This extraordinary book covers the extremely broad subject of electropharmacology defined here as the application of principles and methods of electrochemistry to biological topics associated with the action of drugs. It focuses on the physical principles of the movement of electrical charges across interfaces in pharmacological phenomena. It also covers drugs and the electrical procedures which modify a natural process having an electrochemical basis or component. This outstanding report studies the pharmacologically important properties and effects by electrochemical methods, the electrochemical alternatives or adjuncts to drug therapy and the pharmacology involved. Easy to read and understand, this is an ideal reference for all researchers and practitioners of pharmacology and related fields.

Applied Mechanics Reviews, 1973 **Gaither's Dictionary of Scientific Quotations** Carl C. Gaither, Alma E. Cavazos-Gaither, 2012-01-04 This unprecedented collection of 27 000 quotations is the most comprehensive and carefully researched of its kind, covering all fields of science and mathematics. With this vast compendium, you can readily conceptualize and embrace the written images of scientists, laymen, politicians, novelists, playwrights and poets about humankind's scientific achievements. Approximately 9000 high quality entries have been added to this new edition to provide a rich selection of quotations for the student, the educator and the scientist who would like to introduce a presentation with a relevant quotation that provides perspective and historical background on his subject. Gaither's Dictionary of Scientific Quotations, Second Edition, provides the finest reference source of science quotations for all audiences. The new edition adds greater depth to the number of quotations in the various thematic arrangements and also provides new thematic categories.

Society and the Environment Michael Carolan, 2018-05-15 *Society and the Environment* examines today's environmental controversies within a socio-organizational context. After outlining the contours of pragmatic environmentalism, Carolan considers the pressures that exist where ecology and society collide, such as population growth and its associated increased demands for food and energy. He also investigates how various ecological issues, such as climate change, are affecting our very own personal health. Finally, he drills into the social structural dynamics, including political economy and the international legal system, that create ongoing momentum for environmental ills. This interdisciplinary text features a three-part structure in each chapter that covers fast facts about the issue at hand, examines its wide-ranging implications, and offers balanced consideration of possible real-world solutions. New to this edition are Movement Matters boxes which showcase grassroots movements that have affected legislation. Discussion questions and key terms enhance the text's usefulness, making *Society and the Environment* the perfect learning tool for courses on environmental sociology.

Biological Effects of Ionizing Radiation Marylou Ingram, 1966 *Popular Mechanics*, 2000-01 *Popular Mechanics* inspires, instructs and influences readers to help them master the modern world.

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