

# KEEP IT SIMPLE SCIENCE

**Chemistry Module 5** 

# Chemical Equilibrium **WORKSHEETS**

Worksheet 1	The Basics	Student Name
		4
a) When any chemical reaction reaches an equilibrium in a closed system, what becomes true about the concentration of all the chemical species present?		Why can a dynamic equilibrium NOT occur in an open system?
		5. Write an ionic equation to represent the dynamic
b) Explain why this is so, in the case of a: i) static equilibrium.		equilibrium reached between the solid and the dissolved ions in each of these species, when a solution reaches saturation.
		a) potassium bromide
ii) dynamic equilibrium.		
		b) magnesium chloride
2. What are the thermodynamic reasons why some reactions (eg combustion) are never reversible?		c) iron (III) nitrate
		d) ammonium sulfate
		e) calcium carbonate
3. What is meant by "open" or "closed" chemical systems?		

# **Keep It Simple Science Hsc**

Jim Stamell

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**HSc-Home Science-TB-11 E-R** Kajal Kanjilal, HSc Home Science TB 11 E R

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