

| | | |
|-----|---|-----------------|
| (d) | $\text{C}_5\text{H}_{12}(\text{g}) + 3\text{O}_2(\text{g}) \rightarrow 5\text{C}(\text{s}) + 6\text{H}_2\text{O}(\text{l})$ <p>OR</p> $\text{C}_5\text{H}_{12}(\text{g}) + \frac{11}{2}\text{O}_2(\text{g}) \rightarrow 5\text{CO}(\text{g}) + 6\text{H}_2\text{O}(\text{l})$ | (1 mark) |
| (e) | - Decomposition / breakdown of organic matter ✓ 1 in the absence of oxygen. ✓ 1 | (2 marks) |
| (f) | - Heat a mixture of sodium ethanoate and sodalime ✓ 1 - methane gas is produced and collected over water/ in a syringe/upward delivery ✓ 1 | (2 marks) |
| | | 13 marks |

| | | |
|----|--|--|
| 2. | <p>(a) (i) - Dynamic equilibrium refers to a state of balance/reaction in which forward and backward/reverse reactions are taking place at the same rate. ✓ 1</p> <p>(ii) - The intensity of the orange colour increases. ✓ 1</p> <p>Addition of H^+ ions, sulphuric(VI) acid removes OH^- ions from the equilibrium mixture hence equilibrium shifts to the left. ✓ 1</p> <p>(b) (i) Nitrous acid / HNO_2/ Nitric(III) acid</p> | <p>(1 mark)</p> <p>(2 marks)</p> <p>(1 mark)</p> |
|----|--|--|

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Arshad Iqbal



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Table of Contents Kcse Chemistry Paper 20with Answers

1. Understanding the eBook Kcse Chemistry Paper 20with Answers
 - The Rise of Digital Reading Kcse Chemistry Paper 20with Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Kcse Chemistry Paper 20with Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Kcse Chemistry Paper 20with Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Kcse Chemistry Paper 20with Answers
 - Personalized Recommendations
 - Kcse Chemistry Paper 20with Answers User Reviews and Ratings
 - Kcse Chemistry Paper 20with Answers and Bestseller Lists
5. Accessing Kcse Chemistry Paper 20with Answers Free and Paid eBooks
 - Kcse Chemistry Paper 20with Answers Public Domain eBooks
 - Kcse Chemistry Paper 20with Answers eBook Subscription Services
 - Kcse Chemistry Paper 20with Answers Budget-Friendly Options
6. Navigating Kcse Chemistry Paper 20with Answers eBook Formats

- ePub, PDF, MOBI, and More
 - Kcse Chemistry Paper 20with Answers Compatibility with Devices
 - Kcse Chemistry Paper 20with Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Kcse Chemistry Paper 20with Answers
 - Highlighting and Note-Taking Kcse Chemistry Paper 20with Answers
 - Interactive Elements Kcse Chemistry Paper 20with Answers
 8. Staying Engaged with Kcse Chemistry Paper 20with Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Kcse Chemistry Paper 20with Answers
 9. Balancing eBooks and Physical Books Kcse Chemistry Paper 20with Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Kcse Chemistry Paper 20with Answers
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Kcse Chemistry Paper 20with Answers
 - Setting Reading Goals Kcse Chemistry Paper 20with Answers
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Kcse Chemistry Paper 20with Answers
 - Fact-Checking eBook Content of Kcse Chemistry Paper 20with Answers
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
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