

Population

People are regarded as the greatest assets of the country. This is because people by putting in their labour and expertise play an important part in the development of the economy and of society. For example, minerals such as coal and iron ore become important only when they are put to use by humans. The census of India gives us information about the total population and its various other aspects.

Population Size and Distribution

Population Size

- On March 2001, India's population was 1,028 million which is about 16.7% of the world population.
- Uttar Pradesh, with a population of 166 million, is the most populous state of India. The state accounts for 16% of the total population of the country.
- Sikkim has a population of about 0.5 million people, and Lakshadweep has only 60,000 people.
- It is important to note that nearly half of the population of India lives in five states—Uttar Pradesh, Maharashtra, Bihar, West Bengal and Andhra Pradesh.
- Rajasthan, the largest state in India, has only 5.5% of the total population of the country.

Population Density

- Population density refers to the number of persons occupying or living in per unit of area. India is one of the highly densely populated countries.
- In 2001, the population density of India was 324 persons/sq km.
- While the population density was as high as 904 persons/sq km in West Bengal, it was as low as 13 persons/sq km in Arunachal Pradesh.
- States which have rugged land and unfavorable climatic conditions are sparsely populated. On the contrary, flat plains with moderate climate have high density of population.
- While Sikkim and Mizoram are sparsely populated, West Bengal and Bihar are densely populated.

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web jun 30 2023 major periodic trends include electronegativity ionization energy electron affinity atomic radius melting point and metallic character periodic trends arising from the arrangement of the periodic table provide chemists with an invaluable tool to quickly predict an element's properties

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web jan 28 2019 by mapping specific trends directly onto a blank periodic table the activity uses the periodic table as a foundational aspect of the major trends allowing for students to make connections between placement on the periodic table

and a specific trend

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web you might need periodic table the first two ionization energies for beryllium are shown below be g be g e be g be 2 g e i 1 900 kJ mol i 2 1757 kJ mol which of the following identifies the most probable value for the third ionization energy for be and provides the best justification

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web the purpose of this lab is to introduce students to periodic trends it gives students an opportunity to explore the values of different properties for all elements in the periodic table to determine which properties of the elements follow a periodic trend

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web 1 rank the following elements by increasing atomic radius carbon aluminum oxygen potassium oxygen carbon aluminum potassium 2 rank the following elements by increasing electronegativity sulfur oxygen neon aluminum neon aluminum sulfur oxygen 3 why does fluorine have a higher ionization energy than iodine

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web simulation periodic trends answer key background in this investigation you will examine several periodic trends including atomic radius ionization energy and ionic radius you will

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web 1 how does atomic radius change as you move from left to right across a period explain your answer in terms of the number of protons in the nucleus 2 how does atomic radius change as you move from top to bottom in a group explain your answer in terms of the number of electron shells 3 on the representation of the periodic table below

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web caragabriel preview chemistry test oct 7 11 43 terms misterlominyo preview terms in this set 25 each element has 3 numbers listed under it which value represents the atomic radius the first number below the diagram what are the units for the atomic radius pm picometers 10 negative twelve meters

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web similarities and differences among the elements give rise to so called periodic trends both across rows and within columns of the periodic table recognizing periodic trends in the physical and chemical properties of the elements is key to understanding the full value of the periodic table concepts periodic table periodic trends

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