

| Question | | | Answer | M | Guidance |
|----------|---|-------|---|----------------------|--|
| 6 | a | i | λ is the (de Broglie) wavelength <u>associated with a particle</u> h is (a constant known as) the Planck constant mv is the momentum/mass \times speed of <u>the particle</u> | B1 B1 B1 | allow object/electron for particle accept velocity for speed accept m and v separately but the particle must appear with one of the quantities to score the mark |
| | | ii | electrons passing through a (thin) sheet of graphite are diffracted producing rings/pattern (on a fluorescent screen) | B1 A1 | any suitable situation NOT Au or Al, etc. what is observed + interpretation 1 mark. Do not award the A mark alone unless a plausible situation has been suggested. Young slits type experiments etc score 0/2 |
| | | iii 1 | $eV = \frac{1}{2} mv^2$ $2eV/m = v^2$ so $v = \sqrt{(2 \times 1.6 \times 10^{-19} \times 5.0 \times 10^3 / 9.1 \times 10^{-31})}$ so $v = 1.3 \times 10^6$ (m s ⁻¹) | C1 C1 B1 A0 | allow $eV = 8.0 \times 10^{-17}$ J for 1 mark allow $\sqrt{(2 \times 8.0 \times 10^{-17} / 9.1 \times 10^{-31})}$ or evidence of correct calculation, e.g. $v = 1.325 \times 10^6$ |
| | | iii 2 | $\lambda = h/mv = 6.63 \times 10^{-34} / 1.3 \times 10^6 \times 9.1 \times 10^{-31}$ $\lambda = 5.6 \times 10^{-10}$ (m) | C1 A1 | allow 5.5×10^{-10} if 1.325×10^6 is used |
| | | iii 3 | λ of visible light 5.6×10^{-7} m so power of 10 = 5 or -5 | B1 B1 | ect all 2; accept 4 to 7 $\times 10^{-7}$ m NOT 10^{-7} NOT 10^5 on the answer line |
| | b | | The photoelectric effect | B1 | QWC mark allow exp't description: uv light shone on a zinc plate connected to a gle max 3 from 6 marking points |
| | | | individual photons are absorbed by individual electrons in the metal surface, i.e. one to one interaction. | B1 | |
| | | | Only photons with energies above the work function energy will cause photoelectron emission/idea of threshold frequency | B1 | |
| | | | Hence u-v photons or blue photons will cause photoemission but red photons will not. | B1 | |
| | | | Number of electrons emitted depends on light intensity | B1 | |
| | | | A wave model does not explain instantaneous emission of electrons. | B1 | |
| | | | A wave model does not explain a threshold frequency/wavelength for emission to occur | B1 | allow wave model does not explain no emission however bright the light if energy of photon below work function |
| | | | Total question 6 | 16 | |

G482 Unofficial Mark Scheme

Yijin Wang



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