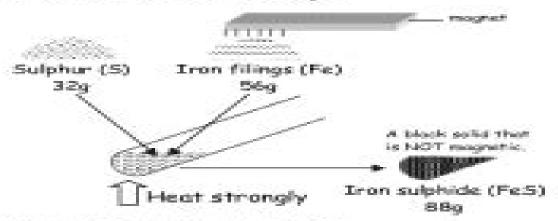
# w.s.st. Chemical reactions.

Name \_\_\_\_\_

All of the different materials around us have been formed by chemical reactions from about one hundred simple elements. The diagram below shows a chemical reaction between the elements iron and sulphur.



This reaction can be shown as a word equation:



The new substance formed is a compound called iron sulphide. It has different properties to the iron and sulphur that it is made from.

Exercise 1 - fill in the missing words in the sentences below.

- The mass of the reactants (starting chemicals) is E \_ \_ \_ \_ to the
  mass of the products (the chemicals that are made).
- 2. The products have different P \_\_\_\_\_ to the reactants.
- 3. During a chemical reaction H \_ \_ \_ is either taken in or given out.
- 4. A chemical change is difficult to R \_\_\_\_\_ (go backwards).

Exercise 2 - Join up each word in the left hand column with its meaning on the right.

ELEMENTS The chemicals that are made.

PRODUCTS The simplest substances.

COMPOUND Starting chemicals

REACTANTS Elements joined together.

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