

ঘের ও কীটপতঙ্গের অংশের নির্ধারিত পদ্ধতির প্রয়োগের

চলকের ঘনমাত্রা,

$$S = \frac{W}{V} = \frac{20}{36.5} = 1$$

$$M = 0.5479 \text{ M}$$

অতএব, HCl চলকের ঘনমাত্রা

$$= 0.5479 \text{ M}$$

এখানে,

সংকেত, $m = 20$ g

HCl এর আপেক্ষিক ভর,

$$M = 36.5 \text{ g/mol}$$

চলকের আয়তন, $V = 1 \text{ L}$

চলকের ঘনমাত্রা $S = 1$

উদাহরণে দ্রবিত X-পদার্থিতে 10 mg এর মধ্যে কয় পরিমাণ দ্রব
যে X-পদার্থিতে 0.5 g 2% $\left(\frac{W}{V}\right)$ HCl দ্রব দিতে কাজ করা হয়েছে।

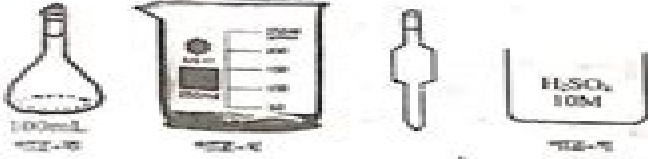
যদিও X-পদার্থটি হলো সেমি-মাইক্রো আনালিটিক্যাল পদ্ধতি ও
Y-পদার্থটি হলো ফার্মা আনালিটিক্যাল পদ্ধতি। উভয় পদ্ধতির
মধ্যে সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব। নিম্নে তা বিশ্লেষণ করা
হলো-

সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি।
এটি বিশ্লেষণের সময় যথেষ্ট রাসায়নিক উপাদান সন্নিবেশিত ও
বিশেষতঃ বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ (0.05
থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে
কাজ করা হয়। ফলে পরিবেশ দূষণ উল্লেখযোগ্যভাবে কমে আসে।
এছাড়া সেমি-মাইক্রো পদ্ধতিতে H_2S এর পরিবর্তে
কোয়েলিনাইট ব্যবহার করা হয় ফলে পরিবেশ দূষণ ঘটে না।



অতএব, কোনো কোনো ক্ষেত্রে কম স্বীকৃত রাসায়নিক দ্রব্য ব্যবহৃত হয়।
ফলে পরিবেশ দূষণের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি
পরিবেশ বান্দব।

চিত্র-১০



১. টাইটার দ্রবক কী?
২. সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব কেন?
৩. চলকের জন্য প্রয়োজ্য হ্যাঙ্গার প্রতীকটি অঙ্কন করে এবং
ব্যাখ্যা করে।
৪. উদাহরণের অংশের দ্রবন থেকে উপযুক্ত কাচ পাত্র ব্যবহার
করে 1M ঘনমাত্রার দ্রবন তৈরি করার প্রক্রিয়া ব্যাখ্যা করে। ৪
ও নং প্রশ্নের উত্তর

৫. বিশ্লেষণের নিমিত্ত বীমের উপর টাইটার স্থাপন করলে বীমের প্রতি
দিকের স্কেল যে দর পাওয়া যায়, তাকে টাইটার দ্রবক বলে।

৬. সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি।
এটি বিশ্লেষণের সময় যথেষ্ট রাসায়নিক উপাদান সন্নিবেশিত ও
বিশেষতঃ বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ
(0.05 থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে
কাজ করা হয়। ফলে পরিবেশ দূষণ উল্লেখযোগ্যভাবে কমে আসে।
অতএব, কোনো কোনো ক্ষেত্রে কম স্বীকৃত রাসায়নিক দ্রব্য ব্যবহৃত হয়।
ফলে পরিবেশ দূষণের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি
পরিবেশ বান্দব।

৭. উদাহরণের দ্রবপট্ট হলো সালফিউরিক এসিড (H_2SO_4) এর 10 M
দ্রব। H_2SO_4 একটি জারক ও ক্ষয়কারী পদার্থ। H_2SO_4 এর 10 M
দ্রব যখন পাত্রে H_2SO_4 দ্রব। পাত্রে H_2SO_4 পদার্থকে ক্ষয়কারী হ্যাঙ্গার
এবং অজৈবিক করা হয়। পাত্রে H_2SO_4 এর অন্য ক্ষয়কারী হ্যাঙ্গার
এবং নিম্নে অঙ্কন করে ব্যাখ্যা করা হলো:

প্রতীক	ক্ষয়কারী (Corrosive)
আয়তন	ক্ষয়কারী রাসায়নিক উপাদান কালে যে ক্ষয় রাসায়নিক দ্রব্যকে দ্রবিত হার সম্পর্কে তথ্য ও অন্যান্য পদার্থের হ্যাঙ্গারের ক্ষতি সঞ্চিত হয়। এতে তথ্য পুঙ্কে হার এবং অন্যান্য পদার্থ পুঙ্কে হার।
উদাহরণ	পাত্রে কঠিন এসিডসলফিউরিক (HCl , HNO_3 , H_2SO_4) পাত্র ক্ষয় (NaOH , KOH , H_2O_2) জারকজীবা সিলিকা প্রতিঃ দ্রবন ইত্যাদি।
সংরক্ষণ পদ্ধতিঃ	- পাত্র এসিডকে পোতলে রাখতে হবে। - পাত্র ক্ষয়কে, ক্ষয় সম্পর্কে বিপাককে পোতলে রাখতে হবে।

৮. উদাহরণের দ্রবপট্ট হলো 10M H_2SO_4 দ্রবন এবং এর থেকে 1M
ঘনমাত্রার দ্রবন প্রস্তুত করতে হবে 1ম ও ২ম চলকের ব্যবহার করতে
হবে।

1ম চলকের দ্রবপট্ট হলো 100 ml জারকজীবা স্কেল ও এর দ্রব
নিম্নে। যেহেতু জারকজীবা স্কেলের জারক 100 ml, তাই
আমাদের 100 ml 1M ঘনমাত্রার H_2SO_4 দ্রবন তৈরি করতে হবে।
দ্রবনের সন্নিবেশ পুঙ্কে হতে জানতে পারি,

$$V_1 S_1 = V_2 S_2$$

$$\text{বা, } V_1 = \frac{V_2 S_2}{S_1}$$

$$= \left(\frac{100 \times 1}{10} \right) \text{ mL}$$

$$= 10 \text{ mL}$$

অতএব, 10ml H_2SO_4 দ্রবনের সঙ্গে পানি যোগ করতে হবে
(100-10) mL = 90 mL
আবার, দ্রবপট্টের সাহায্যে 10 ml দ্রবের ঘনমাত্রার দ্রবন নিম্নে তার
সাথে জারকজীবা স্কেলে 90 ml পানি যোগ করলে আমাদের
কামিত ঘনমাত্রার দ্রবন পাওয়া যাবে।

চিত্র-১১



১. অ্যাসিডজীবা স্কেল কাকে বলে?
২. সোডা দ্রবক K₂ এর ঘন কখনো অসীম হয় না-ব্যাখ্যা
করে।
৩. A স্কেলের হ্যাঙ্গারকে উত্তর করার কৌশল বর্ণনা করে।
৪. B ও C স্কেলের সাথে সম্পর্কিত রাসায়নিক দ্রব্যের
ব্যবহারকালে খান্দা স্বীকৃত হলে কবীর উপর সম্পর্কে
বিশ্লেষণ করে। ৪
ও নং প্রশ্নের উত্তর

৫. অ্যাসিডজীবা স্কেল এক ধরনের রাসায়নিক বা কৃত্রিম দ্রব্য সঞ্চিত
বা দ্রবিত বা উপাদানসমূহকে জারক প্রক্রিয়ায় জারক থেকে তথ্য করে
খান্দার সঞ্চিত করে।

৬. সামান্যতঃ K₂ এর ঘন কখনো অসীম বা শূন্য হতে পারে না।
উদাহরণ ব্যাখ্যা করার জন্য নিম্নের উদাহরণী সীমিতকরণটি বিশ্লেষণ করি-
 $A + B \longrightarrow C + D$

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JG Myers



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