#### Question 2.1:

Fill in the blanks

(a) The volume of a cube of side 1 cm is equal to ....m3

(b) The surface area of a solid cylinder of radius 2.0 cm and height 10.0 cm is equal to ... (mm)2

(c) A vehicle moving with a speed of 18 km h-1 covers....m in 1 s

(d) The relative density of lead is 11.3. Its density is . . . . g cm<sup>-3</sup> or ....kg m<sup>-3</sup>.

#### Answer 2.1:

(a) 1 cm = 
$$\frac{1}{100}$$

Volume of the cube = 1 cm3

But, 
$$1 \text{ cm} 3 = 1 \text{ cm} \times 1 \text{ cm} \times 1 \text{ cm} = \left(\frac{1}{100}\right) \text{m} \times = \left(\frac{1}{100}\right) \text{m} \times = \left(\frac{1}{100}\right) \text{m}$$

$$1 \text{ cm}^3 = 10^{-6} \text{ m}^3$$

Hence, the volume of a cube of side 1 cm is equal to 10-6 m3.

(b) The total surface area of a cylinder of radius r and height h is

 $S = 2\pi r (r + h)$ . Given that, r = 2 cm = 2 × 1 cm = 2 × 10 mm = 20 mm

 $h = 10 \text{ cm} = 10 \times 10 \text{ mm} = 100 \text{ mm} = 15072 = 1.5 \times 10^4 \text{ mm}^2$ 

$$S = 2 \times 3.14 \times 20 \times (20 + 100)$$

(c) Using the conversion,

$$1 \text{ km/h} = \frac{5}{18} \text{ m/s}$$

$$18 \text{ km/h} = 18 \times \frac{5}{18} = 5 \text{ m/s}$$

Therefore, distance can be obtained using the relation: Distance = Speed  $\times$  Time =  $5 \times 1 = 5$  m. Hence, the vehicle covers 5 m in 1 s.

(d) Relative density of a substance is given by the relation,

Density of water = 
$$1 \text{ g/cm}^3$$
  
Density of lead = Relative density of lead × Density of water =  $11.3 \times 1 = 11.3 \text{ g/cm}^3$ 

Again, 
$$1g = \frac{1}{100} kg$$

$$1 \text{ cm}^3 = 10^{-6} \text{ m}^3$$

$$1 \text{ g/cm}^3 = \frac{10^{-3}}{10^{-6}} \text{kg/m}^3 = 10^3 \text{ kg/m}^3$$

$$11.3 \text{ g/cm}^3 = 11.3 \times 10^3 \text{ kg/m}^3$$

#### Question 2.2:

Fill in the blanks by suitable conversion of units

(a) 
$$1 \text{ kg m}^2 \text{ s}^{-2} = \dots \text{ g cm}^2 \text{ s}^{-2}$$

(b) 
$$1 \text{ m} = ..... \text{ly}$$

(c) 
$$3.0 \text{ m s}^{-2} = \dots \text{ km h}^{-2}$$

(d) 
$$G = 6.67 \times 10^{-11} \text{ N m}^2 \text{ (kg)}^{-2} = \dots \text{ (cm)}^3 \text{ s}^{-2} \text{ g}^{-1}$$
.

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# **Grade 11 Physic Term 2 Work 2014**

**Christine D. Tippett, Todd M. Milford** 

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situations are considered some general others with a focus on specific areas of chemistry quantitative problems qualitative reasoning metacognition and resource activation deconstructing the problem solving process an overview of the working memory hypothesis reasoning with the electron pushing formalism scaffolding organic synthesis skills spectroscopy for structural characterization in organic chemistry enzyme kinetics problem solving in the academic chemistry laboratory chemistry problem solving in context team based active learning technology for molecular representations IR spectra simulation and computational quantum chemistry tools The book concludes with methodological and epistemological issues in problem solving research and other perspectives in problem solving in chemistry With a foreword by George Bodner

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