

MARKING SCHEME
CHEMISTRY PRACTICAL
BOKAKE TERM 1 2024

QUESTION 1

TABLE 1

- ✓ complete table CT 1 mk
- ✓ decimal point D 1 mk
- ✓ accuracy A (tied to school value) 1 mk
- ✓ principal of averaging 1 mk
- a) i) Average volume / final accuracy 1 (tied to school value)

NB theoretically expected value = 15.0 cm³

ii) $\frac{1M \times 75 \text{ cm}^3}{250 \text{ cm}^3}$

½ mk

= 0.3M

½ mk

b) $\frac{\text{Ans} \in (b) \text{ above } \times \text{ans} \in (a) \text{ above } \frac{1}{2} \text{ mk}}{1000} = \text{ans}$ **½ mk**



d) Mole ratio = 2:1 **½ mk**

e) Moles of solution W = 2 x ans in a(iii) **½ mk**

f) $\frac{\text{ans} \in b(i) \times 1000 \frac{1}{2} \text{ mk}}{25} = \text{ans}$ **½mk**

g) $\frac{\text{ans} \in (i) \times 600 \frac{1}{2} \text{ mk}}{25} = \text{ans}$ **½ mk**

h) $\frac{8.7}{\text{ans} \in b(iii)} \frac{1}{2} \text{ mk} = \text{RFM OF MOH} \frac{1}{2} \text{ mk}$

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EW Minium



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Revise A2 Chemistry for Salters (OCR) Daniels,2005-04-04 Helps students to pull together key ideas in the course and apply them to exam questions in a fresh context Organised by module to allow readers to quickly access specific information this work provides tips on common pitfalls and advice on approaching exam questions with practice style exam questions for each module along with answers [Index Medicus](#) ,2003 Vols for 1963 include as pt 2 of the Jan issue Medical subject headings

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