

Name: _____

Honors Chem Test: Ch. 7


Multiple Choice—1 point each

Identify the choice that best completes the statement or answers the question.

- How many atoms of fluorine are present in a molecule of carbon tetrafluoride, CF_4 ?
 - 1
 - 2
 - 4
 - 5
- Name the compound Al_2S_3 .
 - aluminum sulfate
 - aluminum sulfur
 - aluminum(II) sulfate
 - aluminum sulfide
- Name the compound CF_4 .
 - calcium fluoride
 - carbon fluoride
 - carbon tetrafluoride
 - mono-carbon quadrafluoride
- Name the compound N_2O_5 .
 - dinitrogen oxide
 - nitrogen trioxide
 - nitric oxide
 - dinitrogen trioxide
- What is the formula for diarsenic trisulfide?
 - As_2S_3
 - AsS_3
 - As_3S_4
 - As_3S_3
- What is the formula for sulfur dichloride?
 - SCl
 - SCl_2
 - S_2Cl
 - S_2Cl_2
- What is the formula for hydrochloric acid?
 - HF
 - HCl
 - HClO
 - H_2CO_3
- The molar mass of an element is the mass of one
 - atom of the element.
 - crystal of the element.
 - gram of the element.
 - mole of the element.
- The molar mass of MgI_2 is
 - the sum of the masses of 1 mol of Mg and 2 mol of I.
 - the sum of the masses of 1 mol of Mg and 1 mol of I.
 - the sum of the masses of 1 atom of Mg and 2 atoms of I.
 - the sum of the masses of 1 atom of Mg and 1 atom of I.

Honors Chemistry Chapter Practice Test

Richard H. Langley



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