

**NEB CLASS 11**  
**MODEL QUESTION**  
**(EFFECTIVE FROM 2077)**

**Grade XI**

**Subject Code : Che. 301**

**Time :3 hours**

**Attempt all questions**

**Subject: Chemistry (Theory)**

**Full marks: 75**

**Group 'A'**

**Circle the best alternative to the following questions.**

**(11×1 = 11)**

1. How many atoms are there in two molecules of water?
  - a. 3
  - b. 4
  - c. 5
  - d. 6
2. What is the number of moles of ammonia gas formed when 0.5 mole of nitrogen gas is reacted with excess of hydrogen gas?
  - a. 0.5
  - b. 1
  - c. 2
  - d. 3
3. Which of the following bonding is responsible for the solubility of ammonia gas in water?
  - a. Hydrogen bonding
  - b. Ionic bonding
  - c. Covalent bonding
  - d. Van der Waals' force
4. What happens when Sulphur dioxide ( $\text{SO}_2$ ) gas is passed through an acidified solution of hydrogen sulfide ( $\text{H}_2\text{S}$ ) gas?
  - a.  $\text{SO}_2$  is oxidized to Sulphur
  - b.  $\text{H}_2\text{S}$  is reduced to Sulphur
  - c.  $\text{SO}_2$  is oxidized to  $\text{H}_2\text{SO}_4$
  - d.  $\text{SO}_2$  is reduced to Sulphur

# Grade 11 Chemistry Paper 2 June 2013

**RJ Alexander**



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