

Name: _____ Date: _____ Period: _____

NOVA: Hunting the Elements

Part 1: Basic Chemistry

Blank Periodic Table

Fill in the parts of this periodic table covered by the documentary.

1 H 1.00794																		2 He 4.002602																			
3 Li 6.941		4 Be 9.012182																		5 B 10.811		6 C 12.011		7 N 14.00643		8 O 15.999		9 F 18.998403		10 Ne 20.1797							
11 Na 22.98976928		12 Mg 24.304																		13 Al 26.9815386		14 Si 28.08558		15 P 30.973762		16 S 32.06		17 Cl 35.453		18 Ar 39.948							
19 K 39.0983		20 Ca 40.078		21 Sc 44.955912		22 Ti 47.88		23 V 50.9415		24 Cr 51.9961		25 Mn 54.938045		26 Fe 55.845		27 Co 58.933195		28 Ni 58.6934		29 Cu 63.546		30 Zn 65.38		31 Ga 69.723		32 Ge 72.64		33 As 74.9216		34 Se 78.96		35 Br 79.904		36 Kr 83.798			
37 Rb 85.4678		38 Sr 87.62		39 Y 88.90584		40 Zr 91.224		41 Nb 92.90638		42 Mo 95.94		43 Tc 98.9062		44 Ru 101.07		45 Rh 102.9055		46 Pd 106.9056		47 Ag 107.8682		48 Cd 112.411		49 In 114.818		50 Sn 118.710		51 Sb 121.757		52 Te 127.6		53 I 126.905		54 Xe 131.29			
55 Cs 132.90545196		56 Ba 137.327		57-70 *		71 Lu 174.967		72 Hf 178.49		73 Ta 180.9479		74 W 183.84		75 Re 186.207		76 Os 190.23		77 Ir 192.222		78 Pt 195.084		79 Au 196.966569		80 Hg 200.59		81 Tl 204.3833		82 Pb 207.2		83 Bi 208.9804		84 Po 209		85 At 210		86 Rn 222	
87 Fr 223		88 Ra 226		89-102 (*) (*)		103 Lr 260		104 Rf 261		105 Db 262		106 Sg 266		107 Bh 264		108 Hs 277		109 Mt 268		110 Uun 288		111 Uuu 289		112 Uub 285		114 Uuq 289											

^a Lanthanide series.

57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb
89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No

" " *Actinidia speciosa*

Gold - Au

1. Write the number of subatomic particles in gold:
 - a. Protons:
 - b. Neutrons:
 - c. Electrons:
2. Give one property of gold.
3. How much gold is in one ton of the mined rock?

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Lei Shi



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